SCHEME OF WORK 2007

CHEMISTRY

FORM FIVE

FIRST TERM -03.01.2007 to 09.03.2007 (10 weeks) 19.03.2007 to 25.05.2007 (10 weeks)

THEME LEARNING AREA ENTERACTION REPORTS CHEMOCALS 1 RATE OF REACTION

No. of week/ Weeks	Learning Oligedbyes	Suggested Learning Activities	Learning Outcomes	Thinking Skills	Scientific Skills	Scientific Atthudes And Noble Values	Remarks/ 'Woaltuilary
	Prusiyoing rate of reaction	Discours: (a) the meaning of rate of reaction. (b) some examples of first reactions. (c) some examples of first reactions. (d) some examples of slow reactions. Discours to identify observable changes to reactions. Discours to identify observable in order to determine the rate of reaction. Carry out an activity involving a reaction between sinc and acid, and plot a graph to determine average rate of reaction at any given time. Carry out problem solving activities involving a resolving activities involving rates of reaction.	A student is able to: state what rate of reaction is. lidentity observable changes to reactants or products for determining rate of reaction. determine average rate of reaction, determine the rate of reaction, determine the rate of reaction at any given time from a graph. solve numerical problems average rate of reaction, tolve numerical problems involving average rate of reaction. solve mumerical problems for a solve involving rate of reaction.	generating ideas relating making inpotheral realising conclusion protoken solving	Observing Inspothershing Communication S Interpreting data	Being hone st and accounts in ne-conding and uslidating data. Being cooperative Being systematic	reactant brotain balas balas product- masil findsin balas rate of reaction balas sobservable change persubahan yang dayar diperhatikan

1

No. of week? Weeks	Learning Objectives	Suggested Learning Activities	Learning Outcomes	Thinking Skills	Scientific Skills	Scientific Attitudes And Noble Values	Pernanks/ Viscabulary
2	1.2 Synthesising declars affecting the rate of reaction	Discouse possible factors affecting the rate of reaction. Design and carry-out activities to investigate factors affecting the rate of reactions, i.e. size of reactions, i.e. size of reactions, i.e. size of reactions and catalyst. Some suggested reactions: (a) a reaction between calcium cartionate. CaCO., and hydrochioris acid. HCI, (b) a reaction between sodium cartionate. Na.5.O., and sulphusic acid. HCS.O., (c) decomposition of hydrogen personice, HCO., in the presence of a catalyst.	A student is able to: design experiment to investigate factors affecting the rate of reaction. give examples of reactions that are affected by size of reactions temperature and catalyst.	generation g ideas relating enabling inpothersi si sanatyring contribution of the cont	predicting predicting date of the second seco	Being honest in necording and validating data Being systematic being cooperative being confident and independent	cuttalydt - in angliën de comporation- penguraian
		Uses computer attimulations to innestigate how the movement and collision of granticles in a reaction are affected by temperature, size of reactant, pressure, concentration and catalyst. Collect and interpret data to explain factors affecting the rate of reaction in the fallowing.	englain how each Sector affects the rate of reaction. describe how Sectors affecting the rate of				

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Table of Contents Federal Scheme Of Work For Sschemistry

- 1. Understanding the eBook Federal Scheme Of Work For Sschemistry
 - The Rise of Digital Reading Federal Scheme Of Work For Sschemistry
 - $\circ\,$ Advantages of eBooks Over Traditional Books
- 2. Identifying Federal Scheme Of Work For Sschemistry
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Federal Scheme Of Work For Sschemistry
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Federal Scheme Of Work For Sschemistry
 - Personalized Recommendations
 - Federal Scheme Of Work For Sschemistry User Reviews and Ratings
 - Federal Scheme Of Work For Sschemistry and Bestseller Lists
- 5. Accessing Federal Scheme Of Work For Sschemistry Free and Paid eBooks
 - Federal Scheme Of Work For Sschemistry Public Domain eBooks
 - Federal Scheme Of Work For Sschemistry eBook Subscription Services

- Federal Scheme Of Work For Sschemistry Budget-Friendly Options
- 6. Navigating Federal Scheme Of Work For Sschemistry eBook Formats
 - ∘ ePub, PDF, MOBI, and More
 - Federal Scheme Of Work For Sschemistry Compatibility with Devices
 - Federal Scheme Of Work For Sschemistry Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Federal Scheme Of Work For Sschemistry
 - Highlighting and Note-Taking Federal Scheme Of Work For Sschemistry
 - Interactive Elements Federal Scheme Of Work For Sschemistry
- 8. Staying Engaged with Federal Scheme Of Work For Sschemistry
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Federal Scheme Of Work For Sschemistry
- 9. Balancing eBooks and Physical Books Federal Scheme Of Work For Sschemistry
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Federal Scheme Of Work For Sschemistry
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Federal Scheme Of Work For Sschemistry
 - Setting Reading Goals Federal Scheme Of Work For Sschemistry
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Federal Scheme Of Work For Sschemistry
 - Fact-Checking eBook Content of Federal Scheme Of Work For Sschemistry
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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