

Name _____ Date _____ Class _____

Modern Chemistry • CHAPTER 6**HOMEWORK 6-5***(pp. 172–173)***VOCABULARY**

Define.

- double bond _____

- triple bond _____
 - How is a triple bond shown? _____
- resonance _____

- resonance structures _____

GRAPHIC ORGANIZER

On a separate sheet of paper, use Table 6-2 on page 173 to create a graph comparing bond length to bond energy. Plot bond length on the x -axis (in pm) and bond energy on the y -axis (in kJ/mol). Depict all the bonds shown in the table on your graph.



What does your graph show about the relationship between bond length and bond energy? Which bonds are stronger—single, double, or triple?

STANDARDIZED TEST PREP

Circle the letter of the best answer.



- The Lewis structure for C_2H_4 is $\begin{array}{c} \text{H} & & \text{H} \\ & \backslash & / \\ & \text{C} = \text{C} \\ & / & \backslash \\ \text{H} & & \text{H} \end{array}$. This Lewis structure shows that
 - the bonds between carbon and hydrogen are double bonds.
 - each hydrogen atom has two valence electrons.
 - the bond between the carbon atoms is a double bond.
 - the bond energies of the carbon-hydrogen bonds are higher than the bond energy of the carbon-carbon bond.
- Which is the correct Lewis structure for HCl?
 - $\text{H} : \ddot{\text{Cl}} :$
 - $\text{H} - \text{Cl}$
 - $^{\cdot}\text{H} - \ddot{\text{Cl}} :$
 - $:\text{H} :: \text{Cl} :$

Modern Chemistry

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Modern Chemistry Homework 5 6

R.F. Price



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Basic Chemistry Concepts and Exercises John Kenkel, 2011-07-08 Chemistry can be a daunting subject for the uninitiated and all too often introductory textbooks do little to make students feel at ease with the complex subject matter. Basic Chemistry Concepts and Exercises brings the wisdom of John Kenkel's more than 35 years of teaching experience to communicate the fundamentals of chemistry in a practical down to earth manner. Using conversational language and logically assembled graphics the book concisely introduces each topic without overwhelming students with unnecessary detail. Example problems and end of chapter questions emphasize repetition of concepts preparing students to become adept at the basics before they progress to an advanced general chemistry course. Enhanced with visualization techniques such as the first chapter's mythical microscope the book clarifies challenging abstract ideas and stimulates curiosity into what can otherwise be an overwhelming topic. Topics discussed in this reader friendly text include Properties and structure of matter Atoms molecules and compounds The Periodic Table Atomic weight formula weights and moles Gases and solutions Chemical equilibrium Acids bases and pH Organic chemicals The appendix contains answers to the homework exercises so students can check their work and receive instant feedback as to whether they have adequately grasped the concepts before moving on to the next section. Designed to help students embrace chemistry not with trepidation but with confidence this solid preparatory text forms a firm foundation for more advanced chemistry training.

Fundamentals of Chemistry: A Modern Introduction (1966) Frank Brescia, 2012-12-02 Fundamentals of Chemistry A Modern Introduction focuses on the formulas processes and methodologies used in the study of chemistry. The book first looks at general and historical remarks definitions of chemical terms and the classification of matter and states of aggregation. The text then discusses gases. Ideal gases pressure of a gas confined by a liquid Avogadro's Law and Graham's Law are described. The book also discusses aggregated states of matter atoms and molecules chemical equations and arithmetic thermochemistry and chemical periodicity. The text also highlights the electronic structures of atoms Quantization of electricity spectra of elements quantization of the energy of an electron associated with nucleus the Rutherford Bohr nuclear theory hydrogen atom and representation of the shapes of atomic orbitals are explained. The text also highlights the types of chemical bonds hydrocarbons and their derivatives intermolecular forces solutions and chemical equilibrium. The book focuses as well on ionic solutions galvanic cells and acids and bases. It also discusses the structure and basicity of hydrides and oxides. The reactivity of hydrides charge of dispersal and basicity effect of anionic charge inductive effect and basicity and preparation of acids are described. The book is a good source of information for readers wanting to study chemistry.

Chemistry in the Modern World Frank L. Wiseman, 1985 Modern Methods of Teaching Chemistry D D Agarwal, 2004 Objective Key Student's Book Pack (Student's Book with Answers with CD-ROM and Class Audio CDs(2)) Annette Capel, Wendy Sharp, 2012-12-28 Objective Key offers students complete official preparation for the Cambridge English Key KET exam. This

is a revised edition of Objective KET Short units offer a variety of lively topics while providing training in exam skills and solid language development Key words sections informed by English Profile highlight the words and meanings A2 level students need to master Spelling Spots and Grammar Extra boxes informed by the Cambridge English Corpus provide help with language points that Key students can find difficult Regular Exam folders and Writing folders provide systematic exam preparation and practice The CD ROM provides activities and games for practice of the target language The CDs include the audio material for the listening exercises

Organic Chemistry Michael B. Smith, 2011-06-29 Based on the premise that many if not most reactions in organic chemistry can be explained by variations of fundamental acid base concepts Organic Chemistry An Acid Base Approach provides a framework for understanding the subject that goes beyond mere memorization The individual steps in many important mechanisms rely on acid base reactions and the ability to see these relationships makes understanding organic chemistry easier Using several techniques to develop a relational understanding this textbook helps students fully grasp the essential concepts at the root of organic chemistry Providing a practical learning experience with numerous opportunities for self testing the book contains Checklists of what students need to know before they begin to study a topic Checklists of concepts to be fully understood before moving to the next subject area Homework problems directly tied to each concept at the end of each chapter Embedded problems with answers throughout the material Experimental details and mechanisms for key reactions The reactions and mechanisms contained in the book describe the most fundamental concepts that are used in industry biological chemistry and biochemistry molecular biology and pharmacy The concepts presented constitute the fundamental basis of life processes making them critical to the study of medicine Reflecting this emphasis most chapters end with a brief section that describes biological applications for each concept This text provides students with the skills to proceed to the next level of study offering a fundamental understanding of acids and bases applied to organic transformations and organic molecules

Teching Of Chemistry: Modern Methods Anjali Khirwadkar, 2004

Education in Modern China R.F. Price, 2014-02-25 First published in 1970 this re issues the revised edition of 1979 This book examines the part played both by tradition and by the Cultural Revolution in the educational system of twentieth century China and explores the apparent reversal of policy which took place since the death of Mao The book discusses the writings of Mao on the nature of man society and knowledge and his ideas on education These are then seen in the context of history philosophy and religion Educational aims and policies are discussed showing how factors such as language geography economics and the social structure created obstacles to reform

Resources in Education , 1990-04

Words for Students of English Holly Deemer Rogerson, Gary Esarey, Linda Schmandt, Dorolyn Smith, 1993 SKILL LEVEL High Beginning Low intermediate Volume 2 consists of 25 units that present basewords with definitions usage examples and exercises Each unit focuses on a specific topic carefully selected for its relevance to students lives so that students can practice new words in meaningful contexts The exercises are flexible and easy to use taking students from

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