

Modern Chemistry Chapter 15

Test Review

- 25 multiple choice
 - definition of self-ionization
 - $[\text{H}_3\text{O}^+][\text{OH}^-] = 1 \times 10^{-14}$
 - definition and formula for calculating $\text{pH} = -\log[\text{H}_3\text{O}^+]$
 - $\text{pH} > 7$ is a base; $\text{pH} < 7$ is an acid; $\text{pH} = 7$ is neutral
 - pH range is normally 0 to 14
 - calculate pH from $[\text{H}_3\text{O}^+]$
 - calculate $[\text{H}_3\text{O}^+]$ from pH
 - definitions of indicators, transition interval, titration, neutralization, endpoint, standard solution
 - acid-base titration monitors pH
 - titration problems using $V_a M_a \text{eq}_a = V_b M_b \text{eq}_b$

Modern Chemistry Chapter 15 Review Acid Base Titration Ph

Michael Brown



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