# 3.1

# MEASUREMENTS AND THEIR UNCERTAINTY

### Section Review

#### Objectives

- Consert measurements to scientific notation.
- Distinguish among the accuracy, precision, and error of a measurement
- Identify the number of significant figures in a measurement and in the result of a calculation

#### Vocabulary

- measurement
- · scientific notation.
- accuracy

- precision.
- accepted value:
- experimental value

- Contract
- percent error
- significant figures.

#### Key Equations

- · Error experimental value accepted value
- Percent error = |error| / accepted value × 100%

#### Part A Completion

Use this completion exercise to check your understanding of the concepts and terms that are introduced in this section. Each blank can be completed with a term, short phase, or number

The \_\_1\_ of a measurement describes how close the

5 from an experimental value. Percent error is calculated 5.
by dividing the 6 of the error by the accepted value and 6.

then multiplying by \_\_\_\_7 \_\_\_\_ 7.

Large and small numbers are more easily handled when 8. expressed in 8. Significant figures in a measurement include 9.

all of the digits that are \_\_\_9 \_\_ plus a last digit that is \_\_10 \_\_\_ 10. \_\_\_\_

# <u>Measurements And Their Uncertainty 3 Pearson</u> <u>Education Answers</u>

**Thomas Griffiths** 

Decoding **Measurements And Their Uncertainty 3 Pearson Education Answers**: Revealing the Captivating Potential of Verbal Expression

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