

**Potentiometry**

Choose the correct answer:

No	Sentence	answer
1	All of the following are related to reference electrode except: a) of known and constant potential b) obeys Nernst equation. c) of high resistance to permit current of electric current. d) insensitive to activity of analyte solution.	C
2	Potential of standard hydrogen electrode is: a) Zero b) related to $H^+$ concentration c) 0.059 PH d) A and B e) All of the above	d
3	The increase in chloride ion concentration in Ag/AgCl electrode leads to: a) increase the potential b) decrease the potential c) no effect d) none of them	b
4	Potential of calomel electrode is ..... proportional to chloride ion concentration. a) directly b) inversely c) equal d) no relation	b
5	In silver/silver chloride electrode, as KCl concentration increases: a) silver ions increases, potential increases. b) silver ions decreases, potential increases. c) silver ions decreases, potential remains the same. d) silver ions decreases, potential decreases.	d
6	In calomel electrode, as KCl concentration increases: a) mercury(I) ions increases, potential increases. b) mercury(I) ions decreases, potential increases. c) mercury(I) ions decreases, potential remains the same. d) mercury(I) ions decreases, potential decreases.	d
7	What is the electrode potential for a half-cell consisting of a cadmium electrode immersed in a solution that is 0.0150 M in $Cd^{2+}$ ( $E^0 = -0.403$ V) a) 0.457 b) 0.403 c) -0.457 d) -0.201	C

# Mcq On Potentiometry

**Mr. Rohit Manglik**



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## **Table of Contents Mcq On Potentiometry**

1. Understanding the eBook Mcq On Potentiometry
  - The Rise of Digital Reading Mcq On Potentiometry
  - Advantages of eBooks Over Traditional Books
2. Identifying Mcq On Potentiometry
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Mcq On Potentiometry
  - User-Friendly Interface
4. Exploring eBook Recommendations from Mcq On Potentiometry
  - Personalized Recommendations
  - Mcq On Potentiometry User Reviews and Ratings
  - Mcq On Potentiometry and Bestseller Lists

5. Accessing Mcq On Potentiometry Free and Paid eBooks
  - Mcq On Potentiometry Public Domain eBooks
  - Mcq On Potentiometry eBook Subscription Services
  - Mcq On Potentiometry Budget-Friendly Options
6. Navigating Mcq On Potentiometry eBook Formats
  - ePub, PDF, MOBI, and More
  - Mcq On Potentiometry Compatibility with Devices
  - Mcq On Potentiometry Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Mcq On Potentiometry
  - Highlighting and Note-Taking Mcq On Potentiometry
  - Interactive Elements Mcq On Potentiometry
8. Staying Engaged with Mcq On Potentiometry
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Mcq On Potentiometry
9. Balancing eBooks and Physical Books Mcq On Potentiometry
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Mcq On Potentiometry
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Mcq On Potentiometry
  - Setting Reading Goals Mcq On Potentiometry
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Mcq On Potentiometry
  - Fact-Checking eBook Content of Mcq On Potentiometry
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

### 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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