

LONDON COLLEGE OF INTERNATIONAL CHEMISTS – 6000 (004)

NAME: _____ COURSE: _____

ENTRY REGISTRATION – APRIL 2000

ICM: 1990 (CONCEPTS IN ORGANIC CHEMISTRY)

ICM 19

Class : 1990-004-000
 Date : 2000 - 2-2000

Days: (1st- _____)

Time : 1:00-2:00

PART – I

20 x 2 = 40 marks

Answer ALL the questions.

1. State three bonding principles.
2. Describe the mechanism of electrophilic aromatic substitution.
3. Write the chemical formulae for $\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_3$.
4. Describe the structure of ethane.
5. What is the relationship between the number of sigma bonds and the number of pi bonds in a molecule?
6. Describe and explain the effect of electronegativity on the chemical reactivity of a molecule.
7. What are the three types of hybridization?
8. What are the three types of hybridization?
9. Describe the effect of electronegativity on the chemical reactivity of a molecule.
10. What are the three types of hybridization?

PART – II

20 x 2 = 40 marks

Answer any FIVE questions.

11. What are the three types of hybridization?
12. Describe the effect of electronegativity on the chemical reactivity of a molecule.
13. Describe the effect of electronegativity on the chemical reactivity of a molecule.
14. Describe the effect of electronegativity on the chemical reactivity of a molecule.
15. Describe the effect of electronegativity on the chemical reactivity of a molecule.
16. Describe the effect of electronegativity on the chemical reactivity of a molecule.
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19. Describe the effect of electronegativity on the chemical reactivity of a molecule.
20. Describe the effect of electronegativity on the chemical reactivity of a molecule.
21. Describe the effect of electronegativity on the chemical reactivity of a molecule.
22. Describe the effect of electronegativity on the chemical reactivity of a molecule.

PART – III

20 x 2 = 40 marks

Answer any FIVE questions.

23. What are the three types of hybridization?
24. Describe the effect of electronegativity on the chemical reactivity of a molecule.
25. Describe the effect of electronegativity on the chemical reactivity of a molecule.
26. Describe the effect of electronegativity on the chemical reactivity of a molecule.
27. Describe the effect of electronegativity on the chemical reactivity of a molecule.
28. Describe the effect of electronegativity on the chemical reactivity of a molecule.
29. Describe the effect of electronegativity on the chemical reactivity of a molecule.
30. Describe the effect of electronegativity on the chemical reactivity of a molecule.
31. Describe the effect of electronegativity on the chemical reactivity of a molecule.
32. Describe the effect of electronegativity on the chemical reactivity of a molecule.

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