

**ICSE Board
Class X Chemistry
Board Paper – 2014**

Time: 1½ hrs

Total Marks: 80

General Instructions:

1. Answers to this paper must be written on the paper provided separately.
2. You will NOT be allowed to write during the first 15 minutes. This time is to be spent in reading the question paper.
The time given at the head of this paper is the time allowed for writing the answers.
This question paper is divided into two sections.
3. **Section I** contains one question with parts (a) to (h); all the eight parts are to be answered.
4. **Section II** contains six questions numbered 2 to 7. You are to answer any four of these questions.
The intended marks of questions or for parts of questions are given in brackets [].

SECTION I (40 Marks)

Attempt all questions from this section.

Question 1

- (a)** Choose the correct answer from the options given below: [10]
- i. Ionisation potential increases over a period from left to right because the
 - A. Atomic radius increases and nuclear charge increases
 - B. Atomic radius decreases and nuclear charge decreases
 - C. Atomic radius increases and nuclear charge decreases
 - D. Atomic radius decreases and nuclear charge increases
 - ii. A compound X consists of only molecules. Hence, X will have
 - A. A crystalline hard structure
 - B. A low melting point and low boiling point
 - C. An ionic bond
 - D. A strong force of attraction between its molecules
 - iii. When fused lead bromide is electrolysed, we observe
 - A. A silver grey deposit at the anode and a reddish brown deposit at the cathode
 - B. A silver grey deposit at the cathode and a reddish brown deposit at the anode
 - C. A silver grey deposit at the cathode and reddish brown fumes at the anode
 - D. Silver grey fumes at the anode and reddish brown fumes at the cathode
 - iv. The main ore used for the extraction of iron is
 - A. Haematite
 - B. Calamine
 - C. Bauxite
 - D. Cryolite

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