

HSC Maharashtra Board question paper: March 2013

Note:

- All questions are compulsory.
- Answer to the two sections are to be written in the same answer book.
- Figure to the right hand side indicate full marks.
- Write balanced chemical equations and draw neat and labelled diagrams wherever necessary.
- Every new question must be started on a new page.
- Use of logarithmic table is allowed.

CHEMISTRY: SECTION – I

Q. 1. Select and write the most appropriate answer from the given alternatives for each sub-question: [7]

- In body centred cubic structure the space occupied is about
(A) 68 % (B) 53 %
(C) 38 % (D) 32 %
- For a gaseous reaction the unit of rate of reaction is
(A) L atm s^{-1} (B) $\text{atm mol}^{-1} \text{s}^{-1}$
(C) atm s^{-1} (D) mol s
- Which of the following compounds contains S – O as well as S – S bonds?
(A) Sulphuric acid (B) Thiosulphuric acid
(C) Sulphurous acid (D) Thiosulphurous acid
- Which of the following solutions shows maximum depression in freezing point?
(A) 0.5 M Li_2SO_4 (B) 1 M NaCl
(C) 0.5 M $\text{Al}_2(\text{SO}_4)_3$ (D) 0.5 M BaCl_2
- For a chemical reaction $\Delta S = -0.035 \text{ kJ/K}$ and $\Delta H = -20 \text{ kJ}$.
(A) 5.14 K (B) 57.14 K
(C) 571.4 K (D) 5714.0 K
- The standard e.m.f of the following cell is 0.463 V
 $\text{Cu} | \text{Cu}_{(0.01)}^{++} || \text{Ag}_{(0.01)}^{+} | \text{Ag}$. If $E_{\text{Ag}}^{\circ} = 0.800 \text{ V}$,
What is the standard potential of Cu electrode?
(A) 1.137 V (B) 0.337 V
(C) 0.463 V (D) -0.463 V
- Fe_2O_3 is reduced to spongy iron near the top of blast furnace by
(A) H_2 (B) CaO
(C) SiO_2 (D) CO

Hsc Unregular Chemistry Question Paper 2012 Bd

Padhraic Smyth



Hsc Unregular Chemistry Question Paper 2012 Bd:

Hsc Unregular Chemistry Question Paper 2012 Bd Book Review: Unveiling the Magic of Language

In an electronic era where connections and knowledge reign supreme, the enchanting power of language has become more apparent than ever. Its power to stir emotions, provoke thought, and instigate transformation is truly remarkable. This extraordinary book, aptly titled "**Hsc Unregular Chemistry Question Paper 2012 Bd**," published by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound affect our existence. Throughout this critique, we shall delve into the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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