

Name _____
stoichiometry

Test next class over

1. In the formation of carbon dioxide from carbon monoxide and oxygen, how many moles of carbon monoxide are needed to react completely with 7.0 moles of oxygen gas?
2. How many moles of oxygen are required to burn 45.8 grams of ethane gas, C_2H_6 , at standard conditions?
3. How many grams of oxygen are produced by the decomposition of 55.0 grams of potassium chlorate, $KClO_3$?
4. A chemist performs the synthesis of sodium chloride from its elements. The chemist begins with 46.55 grams of sodium. How many grams of chlorine are needed?
5. How many grams of water can be prepared from 5.68 moles of hydrogen at standard conditions?
6. Suppose that an excess of propane, C_3H_8 , burns in 62.0 g of O_2 . How many moles of H_2O will be formed?

Grade 1 stoichiometry Practice Test

**Barron's Educational Series, Neil D.
Jespersen, Pamela Kerrigan**



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Reviewing **Grade 1stoichiometry Practice Test**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is actually astonishing. Within the pages of "**Grade 1stoichiometry Practice Test**," an enthralling opus penned by a highly acclaimed wordsmith, readers attempt an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve into the book's central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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