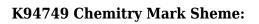
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#### CHEMISTRY THEORY (043)

SECTION A		
1	(c) reacts with Benzenesulphonyl chloride to form a product that is insoluble in alkali	1
2	(b)CH <sub>3</sub> Cl The order followed is this CH <sub>3</sub> I <ch<sub>3Br<ch<sub>3F&lt; CH<sub>3</sub>Cl, though F is most electronegative, the bond length is small as compared to C-Cl. Dipole moment is the product of the charge and the bond length.</ch<sub></ch<sub>	1
3	(a) (i)-(C), (ii)-(B), (iii)-(A)	1
4	(d) 1 =Bromomethane, 2= 2-Bromo-2-methylpropane, 3=2- Bromobutane, 4= 1-Bromobutane	1
	(for visually challenged learners) d. 1-Bromobutane	1
5	(c) the order of reaction is zero as the unit of k is $molL^{-1}s^{-1}$ . Thus half life = $[R]_e$ / $2k$ = $4.62 \times 10^{-2}$ /2 $\times 2.31 \times 10^{-2}$	1
6	(b) Benzoic acid and ethanoic acid  C <sub>6</sub> H <sub>5</sub> COOCOCH <sub>5</sub> H <sub>2</sub> Q C <sub>6</sub> H <sub>5</sub> COOH + CH <sub>3</sub> COOH	1
7	(b) X=[Co(NH <sub>3</sub> ) <sub>4</sub> Cl <sub>2</sub> ] <sup>-</sup> Cl <sup>-</sup> , Y= 1:3	1
8	(b) Cellulose Starch contains only α glucose, sucrose contains α-D-glucose and β-D-fructose glucose, maltose contains α-D-glucose and cellulose is a polymer of β-D-glucose.	1
9	(d) Ti <sup>3+</sup> < Cr <sup>3+</sup> < Fe <sup>2+</sup> < Mn <sup>2+</sup> No. of unpaired electrons: Ti <sup>3+</sup> (1), Cr <sup>3+</sup> (3), Fe <sup>2+</sup> (4) and Mn <sup>2+</sup> (5) Paramagnetism depends on the number of unpaired electrons	1
10	(d) It never goes to completion  First order reaction [R] = [Ro] e <sup>4t</sup> If [R]=0 then	1

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#### **Table of Contents K94749 Chemitry Mark Sheme**

- 1. Understanding the eBook K94749 Chemitry Mark Sheme
  - The Rise of Digital Reading K94749 Chemitry Mark Sheme
  - Advantages of eBooks Over Traditional Books
- 2. Identifying K94749 Chemitry Mark Sheme
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
  - $\circ \ \ Popular \ eBook \ Platforms$
  - Features to Look for in an K94749 Chemitry Mark Sheme
  - User-Friendly Interface
- 4. Exploring eBook Recommendations from K94749 Chemitry Mark Sheme
  - Personalized Recommendations

- K94749 Chemitry Mark Sheme User Reviews and Ratings
- o K94749 Chemitry Mark Sheme and Bestseller Lists
- 5. Accessing K94749 Chemitry Mark Sheme Free and Paid eBooks
  - K94749 Chemitry Mark Sheme Public Domain eBooks
  - K94749 Chemitry Mark Sheme eBook Subscription Services
  - K94749 Chemitry Mark Sheme Budget-Friendly Options
- 6. Navigating K94749 Chemitry Mark Sheme eBook Formats
  - ∘ ePub, PDF, MOBI, and More
  - K94749 Chemitry Mark Sheme Compatibility with Devices
  - K94749 Chemitry Mark Sheme Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of K94749 Chemitry Mark Sheme
  - Highlighting and Note-Taking K94749 Chemitry Mark Sheme
  - Interactive Elements K94749 Chemitry Mark Sheme
- 8. Staying Engaged with K94749 Chemitry Mark Sheme
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers K94749 Chemitry Mark Sheme
- 9. Balancing eBooks and Physical Books K94749 Chemitry Mark Sheme
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection K94749 Chemitry Mark Sheme
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine K94749 Chemitry Mark Sheme
  - Setting Reading Goals K94749 Chemitry Mark Sheme
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of K94749 Chemitry Mark Sheme
  - Fact-Checking eBook Content of K94749 Chemitry Mark Sheme

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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