

HOLT MODERN CHEMISTRY – CHAPTER 1 REVIEW

The following pages contain the bulk (but not all) of the information for the chapter 1 test. Focus on this content, but make sure to review class notes, activities, handouts, questions, etc. If you study this document and NOTHING else, you should at least be able to PASS the test.
****** Test items will be recall, examples, and/or application of this content. ******

OUTCOMES

- 1.1: Distinguish between types of research – basic, applied, and technological development. (F & PK)
- 1.2: Distinguish between the physical and chemical properties of matter. (F & PK)
- 1.2: Distinguish among homogeneous and heterogeneous mixtures, elements, and compounds. (F & PK)
- 1.3: List properties of metals and non-metals. (T & R)
- 2.3: Determine significant figures. (F & PK)

1.1: CHEMISTRY IS A PHYSICAL SCIENCE

- **Vocabulary**
 - **chemistry** – the scientific study of the composition, structure, and properties of matter and the changes that matter undergoes
 - **chemical** – any substance with a defined composition
- **Chapter Highlights**
 - Chemistry is the study of the composition, structure, and properties of matter and the changes that matter undergoes.
 - A chemical is any substance that has a definite composition or is used or produced in a chemical process.
 - Basic research is carried out for the sake of increasing knowledge. Applied research is carried out to solve practical problems. Technological development involves the use of existing knowledge to make life easier or more convenient.

1.2: MATTER AND ITS PROPERTIES

- **Vocabulary**
 - **matter** – anything that has mass and takes up space
 - **mass** -- a measure of the amount of matter in an object
 - **element** – a substance that cannot be separated or broken down into simpler substances by chemical means; all atoms of an element have the same atomic number
 - **atom** – the smallest unit of an element that maintains the chemical properties of that element
 - **compound** -- a substance made up of atoms of two or more different elements joined by chemical bonds
 - **extensive property** -- a property that depends on the extent or size of a system
 - **intensive property** -- a property that does not depend on the amount of matter present, such as pressure, temperature, or density
 - **physical property** -- a characteristic of a substance that does not involve a chemical change, such as density, color, or hardness
 - **physical change** -- a change of matter from one form to another without a change in chemical properties
 - **change of state** -- the change of a substance from one physical state to another
 - **solid** -- the state of matter in which the volume and shape of a substance are fixed
 - **liquid** -- the state of matter that has a definite volume but not a definite shape
 - **gas** -- a form of matter that does not have a definite volume or shape
 - **plasma** -- in physical science, a state of matter that starts as a gas and then becomes ionized; it consists of freemoving ions and electrons and it takes on an electric charge
 - **chemical property** -- a property of matter that describes a substance's ability to participate in chemical reactions
 - **chemical change** -- a change that occurs when one or more substances change into entirely new substances with different properties
 - **chemical reaction** -- the process by which one or more substances change to produce one or more different substances
 - **reactant** -- a substance or molecule that participates in a chemical reaction
 - **product** -- a substance that forms in a chemical reaction

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Table of Contents Holt Modern Chemistry Chapter 7 Review Answer Key

1. Understanding the eBook Holt Modern Chemistry Chapter 7 Review Answer Key
 - The Rise of Digital Reading Holt Modern Chemistry Chapter 7 Review Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Holt Modern Chemistry Chapter 7 Review Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Holt Modern Chemistry Chapter 7 Review Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Holt Modern Chemistry Chapter 7 Review Answer Key
 - Personalized Recommendations
 - Holt Modern Chemistry Chapter 7 Review Answer Key User Reviews and Ratings

- Holt Modern Chemistry Chapter 7 Review Answer Key and Bestseller Lists
- 5. Accessing Holt Modern Chemistry Chapter 7 Review Answer Key Free and Paid eBooks
 - Holt Modern Chemistry Chapter 7 Review Answer Key Public Domain eBooks
 - Holt Modern Chemistry Chapter 7 Review Answer Key eBook Subscription Services
 - Holt Modern Chemistry Chapter 7 Review Answer Key Budget-Friendly Options
- 6. Navigating Holt Modern Chemistry Chapter 7 Review Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Holt Modern Chemistry Chapter 7 Review Answer Key Compatibility with Devices
 - Holt Modern Chemistry Chapter 7 Review Answer Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Holt Modern Chemistry Chapter 7 Review Answer Key
 - Highlighting and Note-Taking Holt Modern Chemistry Chapter 7 Review Answer Key
 - Interactive Elements Holt Modern Chemistry Chapter 7 Review Answer Key
- 8. Staying Engaged with Holt Modern Chemistry Chapter 7 Review Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Holt Modern Chemistry Chapter 7 Review Answer Key
- 9. Balancing eBooks and Physical Books Holt Modern Chemistry Chapter 7 Review Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Holt Modern Chemistry Chapter 7 Review Answer Key
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Holt Modern Chemistry Chapter 7 Review Answer Key
 - Setting Reading Goals Holt Modern Chemistry Chapter 7 Review Answer Key
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Holt Modern Chemistry Chapter 7 Review Answer Key
 - Fact-Checking eBook Content of Holt Modern Chemistry Chapter 7 Review Answer Key
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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downwards answer b is correct the size and direction of the two forces are in balance answer c is incorrect neither the size nor the direction of the two forces are in balance answer d is incorrect

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