

Chapter Test B

Chapter: Chemical Bonding

Match the definition on the left of each statement with the term of the choice that best completes the statement or best answers the question.

1. An ionic bond is formed between atoms of different electronegativities.
a. covalent bond
b. ionic
c. hydrogen
d. metallic
2. A hydrogen bond is a weak attraction between two polar molecules.
a. hydrogen
b. ionic
c. covalent
d. metallic
3. Bond length is the distance between two atoms.
a. bond
b. ionic
c. covalent
d. metallic
4. Bond energy is the amount of energy required to break a bond.
a. bond
b. ionic
c. covalent
d. metallic
5. Bond energy is the amount of energy required to break a bond.
a. bond
b. ionic
c. covalent
d. metallic
6. Bond energy is the amount of energy required to break a bond.
a. bond
b. ionic
c. covalent
d. metallic
7. Bond energy is the amount of energy required to break a bond.
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a. bond
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9. Bond energy is the amount of energy required to break a bond.
a. bond
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d. metallic
10. Bond energy is the amount of energy required to break a bond.
a. bond
b. ionic
c. covalent
d. metallic

Holt Mcdougal Chapter Test B Chapter Chemical Bonding

Katrin Zwirgmaier



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by the body d generating energy for the continuance of the bodily functions correct answer