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alcohols ethanol 2 methylpropan 2 ol butan 2 ol but 2 en 1 ol $\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2\text{OH}$ ii $\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2\text{OH}$ iii $\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2\text{OH}$ iv $\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2\text{OH}$

aldehydes and ketones draw the structures for the following compounds ethanal butanone also known as butan 2 one but the 2 is unnecessary and is often left out 2 methylpropanal

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question 1 basic hydrocarbon nomenclature part a enter the number of carbons indicated by each of the following root names in the designated answer box part b the line formula for a branched alkane is shown below part c the line formula for another

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dec 16 2021 so the complete name is 5 bromo 7 chloro 6 hydroxy 2 2 5 trimethyl 7 octen 4 one 4 it is not difficult to find the parent structure for this compound which is a cyclic alcohol so the last name is cyclopropanol the naming of the substituent with the benzene ring is

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acetaldehyde acetone answer a the iupac name of CH_3CHO is ethanal q2 the iupac name of $\text{CH}_3\text{OCH}_2\text{CH}_2\text{CH}_3$ is methoxy ethane

ethoxy ethane methyl ethane ethyl ethane

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1 carbon methane 2 carbons ethane 3 carbons propane 4 carbons butane 5 carbons pentane 6 carbons hexane 7 carbons

heptane 8 carbons octane 9 carbons nonane 10 carbons decane while many of these names share a greek root with