

MATHEMATICS
GRADE 8
SBA: ASSIGNMENT

MARCH 2021
50 MARKS
1 HOUR

Question 1

MARCH 2021						
Sun	Mon	Tue	Wed	Thu	Fri	Sat
	1	2	3	4	5	6
7	8	9	10	11	12	13
14	15	16	17	18	19	20
21	22	23	24	25	26	27
28	29	30	31			

Use the calendar to answer the following questions.

- 1.1 Write down any TWO prime numbers in the calendar. (2)
- 1.2 List the factors of 24 in the calendar. (2)
- 1.3 Write down the multiples of 9 listed in the calendar. (2)
- 1.4 Look at Sundays' dates: 7; 14; 21; 28.
These numbers are (factors/multiples) of 7.
- Choose the correct answer and write it down on your answer paper. (1)
- [7]

Question 2

- 2.1 Decompose 36 as a product of prime factors. (Use the ladder method) (2)
- 2.2 Determine the HCF of 36 and 54. (4)
- 2.3 Determine the LCM of 24 and 63. (4)
- [10]

Question 3

- 3.1 Write the ratio 6:24 in the simplest form. (2)
- 3.2 Mrs Reed shares Smarties between Mrs Schwartz and Mrs Lentz in the ratio of 3:5. How many Smarties does Mrs Lentz get, if there are 72 Smarties in the box? (4)
- 3.3 You are given R 150 pocket money each month. You negotiated with your mom to increase the amount in the ratio of 6:5.

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Arturo Cuomo



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a net ionic equation is the most accurate representation of the actual chemical process that occurs writing these equations requires a familiarity with solubility rules acid base reactivity

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apr 25 2017 this is understood when you consider the two chemicals appear unchanged before and after the reaction therefore the net ionic equation is $2\text{Cl}^- + 2\text{Ag}^+$

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ions that remain essentially unchanged during a reaction are called spectator ions this means you can ignore them when you write the ionic equation you only need to model how the solid

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the first step to writing a net ionic equation is balancing the chemical equation present let s use the reaction between sodium chloride and silver nitrate as an example next we write the

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highlights learning objectives by the end of this section you will be able to derive chemical equations from narrative descriptions of chemical reactions write and balance chemical

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the net ionic equation is the chemical equation that shows only those elements compounds and ions that are directly involved in the chemical reaction notice that in writing the net ionic

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this net ionic equation indicates that solid silver chloride may be produced from dissolved chloride and silver i ions regardless of the source of these ions these molecular and

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write the overall chemical equation the complete ionic equation and the net ionic equation for the reaction of aqueous silver fluoride with aqueous sodium phosphate to give solid silver

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the complete ionic equation for this reaction is as follows equation 4 13 2ag aq 2f aq 2nh₄ aq cr₂o₇²⁻ aq ag₂cr₂o₇ s 2nh₄ aq 2f aq because two nh₄ aq

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