

Monosubstituted Alkenes and Compounds

Write the following with structural formulas

Take good notes that compare the structure family tree

10/10/20

In these compounds, which are treated as simple molecules, alkenes contain one double bond, and alkynes two triple bonds. In alkenes, alkynes and alkanes, the atoms are joined in a chain of single bonds. Find the structural formula for propene and the number of atoms in propene.

Alkene	Alkyne	Alkane	Alkene	Alkyne	Alkane	Alkene	Alkyne	Alkane	Alkene	Alkyne	Alkane
<u>C₂H₄</u>	<u>C₂H₂</u>	<u>C₂H₆</u>	<u>C₃H₆</u>	<u>C₃H₄</u>	<u>C₃H₈</u>	<u>C₄H₈</u>	<u>C₄H₆</u>	<u>C₄H₁₀</u>	<u>C₅H₁₀</u>	<u>C₅H₈</u>	<u>C₅H₁₂</u>

Write the formula of these monosubstituted compounds:

propene C₃H₆

butyne C₄H₆

butene C₄H₈

C₄H₆

pentene

C₅H₁₀

C₅H₈

butyne C₄H₆

C₄H₆

pentene C₅H₁₀

C₅H₈

propene C₃H₆

butyne C₄H₆

butene C₄H₈

butyne C₄H₆

C₄H₆

Alkenes, Alkynes, and Alkanes: Family Tree of Monosubstituted Compounds

Organic compounds are made of carbon and hydrogen atoms. Alkenes have double bonds, alkynes have triple bonds, and alkanes have single bonds.

Alkenes

The number of carbon atoms in alkenes is 2 or more.

Alkenes are unsaturated hydrocarbons. They have double bonds. The number of carbon atoms in alkenes is 2 or more.

Alkynes are unsaturated hydrocarbons. They have triple bonds. The number of carbon atoms in alkynes is 2 or more.

Alkanes are saturated hydrocarbons. They have single bonds. The number of carbon atoms in alkanes is 1 or more.

Alkenes are unsaturated hydrocarbons. They have double bonds. The number of carbon atoms in alkenes is 2 or more.

Alkynes are unsaturated hydrocarbons. They have triple bonds. The number of carbon atoms in alkynes is 2 or more.

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Patrick Vollmar



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