

4.2 Consider the equations for reactions I to III below.

A and B represent organic compounds that are POSITIONAL
X is an inorganic product.

I	$\text{CH}_3\text{CH}_2\text{CHCHCH}_3 + \text{HCl} \rightarrow \text{A} + \text{B}$
II	$\text{A} \xrightarrow[\Delta]{\text{H}_2\text{O}} \text{CH}_3\text{CH}_2\text{CH}_2\text{CH}(\text{OH})\text{CH}_3 + \text{X}$
III	$\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}(\text{OH})\text{CH}_3 \rightarrow \text{CH}_3\text{CH}_2\text{CHCHCH}_3 + \text{H}_2\text{O}$

Write down the:

4.2.1 Definition of *positional isomers* (2)

4.2.2 Type of reaction represented by reaction I (1)

4.2.3 STRUCTURAL formula of compound B (3)

4.2.4 Formula of X (1)

...

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PHYSICAL

SCIENCES P.2

QN 4

alkane

HCl

HCl

H₂SO₄

(are they PO...)

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1. Understanding the eBook Memo September 2013 Physical Sciences Paper 2
 - The Rise of Digital Reading Memo September 2013 Physical Sciences Paper 2
 - Advantages of eBooks Over Traditional Books
2. Identifying Memo September 2013 Physical Sciences Paper 2
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Memo September 2013 Physical Sciences Paper 2
 - User-Friendly Interface
4. Exploring eBook Recommendations from Memo September 2013 Physical Sciences Paper 2
 - Personalized Recommendations
 - Memo September 2013 Physical Sciences Paper 2 User Reviews and Ratings
 - Memo September 2013 Physical Sciences Paper 2 and Bestseller Lists
5. Accessing Memo September 2013 Physical Sciences Paper 2 Free and Paid eBooks
 - Memo September 2013 Physical Sciences Paper 2 Public Domain eBooks
 - Memo September 2013 Physical Sciences Paper 2 eBook Subscription Services
 - Memo September 2013 Physical Sciences Paper 2 Budget-Friendly Options

6. Navigating Memo September 2013 Physical Sciences Paper 2 eBook Formats
 - ePub, PDF, MOBI, and More
 - Memo September 2013 Physical Sciences Paper 2 Compatibility with Devices
 - Memo September 2013 Physical Sciences Paper 2 Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Memo September 2013 Physical Sciences Paper 2
 - Highlighting and Note-Taking Memo September 2013 Physical Sciences Paper 2
 - Interactive Elements Memo September 2013 Physical Sciences Paper 2
8. Staying Engaged with Memo September 2013 Physical Sciences Paper 2
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Memo September 2013 Physical Sciences Paper 2
9. Balancing eBooks and Physical Books Memo September 2013 Physical Sciences Paper 2
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Memo September 2013 Physical Sciences Paper 2
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Memo September 2013 Physical Sciences Paper 2
 - Setting Reading Goals Memo September 2013 Physical Sciences Paper 2
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Memo September 2013 Physical Sciences Paper 2
 - Fact-Checking eBook Content of Memo September 2013 Physical Sciences Paper 2
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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