



# Heidelberg Gto 46service Manual

**RD Boyd**



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nh<sub>3</sub> 5 00 l n<sub>2</sub> n<sub>2</sub> 1 l 13 l cl<sub>2</sub> x n<sub>2</sub> 3 l cl<sub>2</sub> 4 3 l n<sub>2</sub> answer 4 3 x 103 ml n<sub>2</sub> b how many grams of chlorine must react to produce 16 l of nitrogen gas at 1 2 atm and 23oc nn<sub>2</sub> 1 2atm x 16l

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h 2 s o 4 n a o h n a 2 s o 4 h 2 o if it takes 27 4 ml of 0 768 m naoh n a o h to titrate 16 7 ml of h<sub>2</sub>so<sub>4</sub> h 2 s o 4 what is the concentration of the h<sub>2</sub>so<sub>4</sub> h 2 s o 4 solution hint balance the equation first q2

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