

Calorimetry: Heat of Neutralization

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Purpose: Determine the calorimeter constant for a calorimeter. Determine the enthalpy of neutralization for the reaction of a strong acid and a strong base.

Background: When a system absorbs heat (q) its temperature changes by ΔT . Experimental measurements demonstrate that the heat absorbed by a system and its corresponding temperature change are directly proportional. $q = C \times \Delta T$. C is called the **heat capacity**, which is often defined as the quantity of heat required to change its temperature by 1°C .

$$C = q / \Delta T = \text{J} / ^\circ\text{C}$$

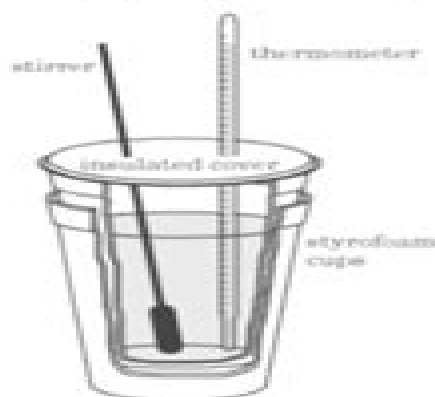
Heat capacity is an extensive property; it depends on the amount of matter being heated. The measure of intrinsic capacity of a substance to absorb heat is called **specific heat capacity** (C_s), the amount of heat required to raise the temperature of 1 gram of the substance by 1°C .

$$C_s = q / (\Delta T \times m)$$

The C_s of H_2O is $4.184 \text{ J/g}^\circ\text{C}$. For the case where all the heat is used to heat water, the amount of heat is calculated by the relation

$$q = C_s \times m_{\text{water}} \times \Delta T$$

The enthalpy change for a chemical reaction, ΔH_{rxn} is also called the **enthalpy of reaction** or **heat of reaction**, which is also an extensive property. For many aqueous reactions, ΔH_{rxn} can be measured simply using a coffee-cup calorimeter



Heat Of Neutralization Lab Manual

Jo A. Beran



Heat Of Neutralization Lab Manual:

Reviewing **Heat Of Neutralization Lab Manual**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is really astonishing. Within the pages of "**Heat Of Neutralization Lab Manual**," an enthralling opus penned by a very acclaimed wordsmith, readers embark on an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve in to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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