PHYSICAL SCIENCES - GRADE 10

1 Revision of matter (Grade 9)

1.1 Matter and classification

- Matter is made up of particles (atoms or molecules) and it is the properties of the atoms or molecules that determine the characteristics and reactivity of that matter.
- The properties of matter include: strength, density; melting and boiling points, whether the material is brittle, malleable or ductile, whether it is magnetic or not; and its electrical and thermal conductivity.
- Elements and compounds are classified as pure because they contain particles that are all the same. Substances can be classified as pure – elements and compounds

- or mixtures. Pure substances contain particles that are all the same. Elements and compounds can be represented by symbols and formulae. Some of the names, symbols or formulae of a number of common elements and compounds that you should have learnt are the following:

- Elements: S for sulphur, C for carbon, P for phosphorus (note the spelling).
 H for hydrogen, O for oxygen, He for helium, N for nitrogen, F for fluorine,
 Mg for magnesium, Ca for calcium, Zn for zinc.
- Compounds: H₂O for water, H₂SO₄ for sulphuric acid; HCl for hydrochloric acid; NaCl for sodium chloride (table salt); CaCO for calcium carbonate; NNO₃ for

potassium nitrate. Na₁PO₄ for sodium phosphate.

- · You will notice that the naming of compounds follows these rules:
 - The metal or hydrogen is always named first, then the non-metal (potassium iodide: Ki, hydrogen chloride: HCl (also called hydrochloric acid)).
 - The name of a non-metal ion of an element always ends in -ide (for instance, sodium chloride: NaCl).
 - The name of a non-metal complex ion (more than one element in the ion)
 usually ends in -ate or -ite (for instance, copper sulphate: CuSO_G potassium
 sulphite: K₂SO₃). An exception to this is hydroxide (for instance, sodium
 hydroxide: NaOH).
- Mixtures consist of different substances, and can be homogeneous (have the same 'look' throughout) or heterogeneous (you can see the different substances). They can be separated in various ways, for instance filtering, sieving, chrom at ography, distilling or dissolving one constituent.
- Some of the properties of metals are that they conduct electricity and heat, and they are malleable and ductile.
- Non-metals are non-conductors of electricity and heat. They are gases, liquids or brittle solids, and bond with each other.
- Metalloids are elements on the border between metals and non-metals, and they have both metal and non-metal properties.

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Guide to Information Sources in the Physical Sciences David Stern, 2000-06-15 This bibliographic guide offers users a basic overview of the current trends and the best most important and most up to date paper and electronic information resources in the field of physics The author has selectively chosen and succinctly annotated a list of hundreds of major tools used by physical scientists and researchers including bibliographic sources abstracting and indexing databases journals books online sources and other subject specific non bibliographic tools Stern also provides information on grants personal bibliographic database tools document delivery copyright and reserves In addition he discusses future developments directions and trends in the field and in the concluding chapter he outlines the history and developments of the physics Designed to help students new researchers in the field of physics and working physicists in need of additional information resources outside their normal field of study this is an invaluable reference research and collectio **Teaching Middle School Science** Smithsonian Institution, National Academy of Engineering, National Science Resources Center of the National Academy of Sciences, Institute of Medicine, 1998-04-30 With age appropriate inquiry centered curriculum materials and sound teaching practices middle school science can capture the interest and energy of adolescent students and expand their understanding of the world around them Resources for Teaching Middle School Science developed by the National Science Resources Center NSRC is a valuable tool for identifying and selecting effective science curriculum materials that will engage students in grades 6 through 8 The volume describes more than 400 curriculum titles that are aligned with the National Science Education Standards This completely new guide follows on the success of Resources for Teaching Elementary School Science the first in the NSRC series of annotated guides to hands on inquiry centered curriculum materials and other resources for science teachers The curriculum materials in the new guide are grouped in five chapters by scientific area Physical Science Life Science Environmental Science Earth and Space Science and Multidisciplinary and Applied Science They are also grouped by type core materials supplementary units and science activity books Each annotation of curriculum material includes a recommended grade level a description of the activities involved and of what students can be expected to learn a list of accompanying materials a reading level and ordering information The curriculum materials included in this book were selected by panels of teachers and scientists using evaluation criteria developed for the guide The criteria reflect and incorporate goals and principles of the National Science Education Standards The annotations designate the specific content standards on which these curriculum pieces focus In addition to the curriculum chapters the quide contains six chapters of diverse resources that are directly relevant to middle school science Among these is a chapter on educational software and multimedia programs chapters on books about science and teaching directories and guides to science trade books and periodicals for teachers and students Another section features institutional resources One chapter lists about 600 science centers museums and zoos where teachers can take middle school students for interactive science experiences Another chapter describes nearly 140 professional associations and U S government agencies that offer resources and assistance Authoritative extensive and thoroughly indexed and the only guide of its kind Resources for Teaching Middle School Science will be the most used book on the shelf for science teachers school administrators teacher trainers science curriculum specialists advocates of hands on science teaching and concerned parents Resources in Education ,1998 Physical Science Teacher's Desk Reference Prentice-Hall in Education ,1972 The Outlines of Literature, English and American Truman Jay Backus, 1897 Staff.1993 Prentice Hall Physical Science Concepts in Action Program Planner National Chemistry Physics Earth Science, 2003-11 Prentice Hall Physical Science Concepts in Action helps students make the important connection between the science they read and what they experience every day Relevant content lively explorations and a wealth of hands on activities take students understanding of science beyond the page and into the world around them Now includes even more technology tools and activities to support differentiated instruction Physics, the Human Adventure Gerald James Holton, Stephen G. Brush, 2001 Of Some Systematic Materials Analysis J. H. Richardson, R. V. Peterson, 2013-10-22 Trigonometric Relations Vector Algebra Systematic Materials Analysis Volume III presents brief discussions on a broad range of instrumental methods and approaches that will yield the desired information about a given material The book discusses the selection of analytical methods on the bases of specimen limitations and information desired The chapters on specific instruments briefly outline the theories of operation and describe the capability of the methods for qualitative and quantitative measurements of chemical composition structure and texture as applicable The commercial instruments and techniques discussed include arc spark laser plasmas flame photometry gas analysis techniques combustion methods gas chromatography and ion scattering spectrometry The Mossbauer spectrometry optical microscopy x ray diffraction x ray fluorescence and absorption spectrometry are also encompassed Materials analyst materials scientist chemists and engineers will find the book invaluable

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