

Unit 7: Chemical Formulas and Compounds Guided Notes

I. Ch. 7.1: Naming Ionic Compounds

A. Warm Up:

1. Distinguish between a cation and an anion.
2. Which elements tend to form cations? anions?
3. Using a periodic table, predict the charge ion each of the following elements form:
 - a) calcium
 - b) fluorine
 - c) sulfur
 - d) aluminum

B. Writing Formulas and Naming Chemical Compounds

1. There are two types of chemical compounds:
 - a) ionic compounds
 - b) covalent (molecular) compounds

C. Ionic Compounds are composed of a cation and an anion.

1. cations – positively charged ion (_____)
2. anions – negatively charged ion (_____)
3. Monoatomic ions
 - a) Cations
 - (1) For monatomic cations, the name of the ion is the name of the element.
 - (2) examples:
 - (a) Na^+ = _____
 - (b) Mg^{2+} = _____
 - (c) Al^{3+} = _____
 - b) Some metals are able to form more than one stable ion (_____)
 - (1) usually transition elements
 - (2) example: _____
 - (3) forms Cu^{2+} and Cu^+
 - (4) use Roman numerals to distinguish charge
 - (5) copper (II) and copper (I)

Guided And Study Chemical Names Formulas

M Carnoy



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General, Organic, and Biochemistry Study Guide Marcia L. Gillette, Ira Blei, Wendy Gloffke, George Odian, 2006-01-27
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Scientific Information Notes ,1961 **Hazmat Chemistry Study Guide (Second Edition)** Jill Meryl Levy, 2011

Guide to Sources for Agricultural and Biological Research J. Richard Blanchard, Lois Farrell, 2023-07-28 **Study Guide to Accompany Basics for Chemistry** Martha Mackin, 2012-12-02 Study Guide to Accompany Basics for Chemistry is an 18 chapter text designed to be used with Basics for Chemistry textbook Each chapter contains Overview Topical Outline Skills and Common Mistakes which are all keyed to the textbook for easy cross reference The Overview section summarizes the content of the chapter and includes a comprehensive listing of terms a summary of general concepts and a list of numerical exercises while the Topical Outline provides the subtopic heads that carry the corresponding chapter and section numbers as they appear in the textbook The Fill in Multiple Choice are two sets of questions that include every concept and numerical exercise introduced in the chapter and the Skills section provides developed exercises to apply the new concepts in the chapter to particular examples The Common Mistakes section is designed to help avoid some of the errors that students make in their effort to learn chemistry while the Practical Test section includes matching and multiple choice questions that comprehensively cover almost every concept and numerical problem in the chapter After briefly dealing with an overview of chemistry this book goes on exploring the concept of matter energy measurement problem solving atom periodic table and chemical bonding These topics are followed by discussions on writing names and formulas of compounds chemical formulas and the mole chemical reactions calculations based on equations gases and the properties of a liquid The remaining chapters examine the solutions acids bases salts oxidation reduction reactions electrochemistry chemical kinetics and equilibrium and nuclear organic and biological chemistry This study guide will be of great value to chemistry teachers and students **Practical Guide to Chemical Safety Testing** Derek J. Knight, Mike B. Thomas, 2003 Annotation This book describes the different tests that must be performed on new chemicals and other materials to demonstrate to the regulatory authorities that they are safe for use Tests vary from physico chemical measuring properties such as melting point and density through genetic toxicity studies to mammalian toxicology and studies to investigate effects on the environment This book describes clearly the process of obtaining approval for use in a variety of global regions and across different applications It also explains why different tests are performed and the implications of the results **SSC Constable Exam Guide 2020** Arihant Experts, 2020-01-11 *Chemistry, Student Study Guide* James E. Brady, Fred Senese, 2008-01-28 The image on the front cover depicts a carbon nanotube emerging from a glowing plasma of hydrogen and carbon as it forms

around particles of a metal catalyst Carbon nanotubes are a recently discovered allotrope of carbon Three other allotropes of carbon buckyballs graphite and diamond are illustrated at the left as is the molecule methane CH_4 from which nanotubes and buckyballs can be made The element carbon forms an amazing number of compounds with structures that follow from simple methane found in natural gas to the complex macromolecules that serve as the basis of life on our planet The study of chemistry also follows from the simple to the more complex and the strength of this text is that it enables students with varied backgrounds to proceed together to significant levels of achievement

A Practical Guide to Using Glycomics Databases Kiyoko F. Aoki-Kinoshita, 2016-12-06 This book provides glycoscientists with a handbook of useful databases that can be applied to glycoscience research Although many databases are now publicly available one of the hurdles for their users is the learning curve required to effectively utilize those databases Therefore this book not only describes the existing databases but also provides tips on how to obtain the target data That is because many databases provide a variety of data that could be obtained from different perspectives each chapter provides users with potential biological questions that can be answered by a particular database and step by step instructions with figures on how to obtain that data Troubleshooting tips are also provided to aid users encountering problems that can be predicted when using these databases Moreover contact information for each database is provided in case unexpected issues arise

A Practical Guide to Understanding, Managing, and Reviewing Environmental Risk Assessment Reports Sally L. Benjamin, David A. Belluck, 2001-02-21 A Practical Guide to Understanding Managing and Reviewing Environmental Risk Assessment Reports provides team leaders and team members with a strategy for developing the elements of risk assessment into a readable and beneficial report The authors believe that successful management of the risk assessment team is a key factor is quality report

Study Guide/Selected Solutions Manual Julie R. Frentrop, 2002-06 Contains a brief overview of every chapter review of skills self tests and the answers and detailed solutions to all odd numbered end of chapter problems in the text book

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A Guide to Safe Material and Chemical Handling Nicholas P. Cheremisinoff, Anton Davletshin, 2010-04-27 There have been many volumes written that claim to be the most comprehensive compendium or handbook on chemical data These wieldy volumes are often too big and extraneous to be useful to the practicing engineer This new volume aims to be the most useful go to volume for the working

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Richard Paul, Linda Elder, 2019-06-01 The Thinker's Guide to Scientific Thinking focuses on developing the intellectual skills inherent in the well cultivated practice of every area of scientific research and study It helps students and practicing scientists come to reason within the logic of science and to see the field as a cohesive whole From astronomers to zoologists and physicists to chemists skilled scientists use careful analysis to question data test theories draw logical conclusions and propose feasible solutions Students in science courses and scientists themselves will find their analytical abilities enhanced by the engaging framework of inquiry set forth

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