

SECTION 7.2 Erosion and Deposition

In your textbook, read about erosion and deposition.

For each item in Column A, write the letter of the matching item in Column B.

	Column A	Column B
_____ F _____	1. The final stage of the erosional process in which materials are dropped in another location	a. slope
_____ E _____	2. The force that tends to pull all materials downhill	b. ocean waves
_____ A _____	3. The steeper the _____, the greater the potential for flowing water to erode earth materials.	c. wind
_____ B _____	4. Coastal areas undergo erosion by _____ and wind.	d. glaciers
_____ D _____	5. Erode by scraping, gouging, and picking up large rocks and debris piles	e. gravity
_____ C _____	6. A major erosional agent in areas with limited precipitation and high temperatures	f. deposition

Answer the following questions.

7. Give two examples of how plants and animals move Earth's surface materials from one place to another as they carry on their life processes.

Moles would dig up the soil in the ground and bury it in another place. Plants, when you garden you are picking up the subsoil and moving it and using it to plant another plant.

8. Explain rill erosion and how it differs from gully erosion.

Rill erosion is the erosion by running water in small channels and gully erosion is when a channel becomes deep and wide.

9. Describe the formation of barrier islands.

Forms when the constant movements of water wears away shorelines.

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This book delves into Glencoe Integrated Science Chapter 10 Answer Key. Glencoe Integrated Science Chapter 10 Answer Key is a vital topic that needs to be grasped by everyone, ranging from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Glencoe Integrated Science Chapter 10 Answer Key, encompassing both the fundamentals and more intricate discussions.

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 $\text{BaSO}_4 + \text{S} + \text{Ba} + 2\text{A} + \text{Q} + \text{SO}_4 + 2\text{A} + \text{Q} + \text{X} + \text{Ag}_2 + \text{SO}_4 + 2\text{S} + 2\text{Ag} + \text{A} + \text{Q} + \text{SO}_4 + 2\text{A} + \text{Q} + \text{X} + \text{Al} + \text{OH} + 3\text{S} + \text{Al} + 3\text{A} + \text{Q} + 3\text{OH} + \text{A} + \text{Q} + \text{X} + \text{Pb} + \text{OH} + \text{Cl} + \text{S} + \text{Pb} + 2\text{A} + \text{Q} + \text{OH} + \text{A} + \text{Q} + \text{Cl} + \text{A} + \text{Q} + \text{X}$

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 $\text{Ca}^{2+}(\text{aq}) + 2\text{OH}^{-}(\text{aq}) + 2\text{H}^{+}(\text{aq}) + \text{CO}_3^{2-}(\text{aq}) \rightarrow \text{Ca}^{2+}(\text{aq}) + \text{CO}_3^{2-}(\text{aq}) + \text{H}_2\text{O}(\text{l})$

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