



Electron Configuration

Example 1:
Small atoms

Write the following:

- On the periodic table: 1s, noble gasses,
- Elements 19-36 (periodic table)
- Lanthanum



Electron configurations are arranged in **blocks**, **columns**, and **rows**. These levels of organization are changed by the shape of the orbitals. Electrons fill orbitals in order. The orbitals are labeled with letters (s, p, d, f) and with a number (1 through 7) indicating the energy level.

Example: These two configurations are found in elements with atomic numbers 1 through 18:

1. Example: This **1s²2s²2p⁶3s²3p⁶4s²3d⁶** electron configuration shows the ground state configuration. Which energy is **lowest**, the orbitals are arranged from lowest energy on the bottom left to highest energy on the top right. What does it tell us about the order of filling of orbitals? (What is the order of filling?)

What does it tell us about the configuration of hydrogen? **1s¹**

2. Example: What does it tell us about helium and another element on the far right? The configuration is **1s²2s²2p⁶3s²3p⁶4s²3d¹⁰4p⁶**. What is the order of filling of orbitals?

1s²2s²2p⁶3s²3p⁶4s²3d¹⁰4p⁶

The **Pauli exclusion principle** states that no two electrons can have the same set of quantum numbers.

3. Write your own: Helium. Write the electron configuration of helium? **1s²**

4. Example: What does it tell us about the configuration of the elements hydrogen, helium, and lithium? What **does** it tell us about your work, lithium? Its configuration is **1s²2s²2p⁶3s¹**.

Hydrogen: **1s¹**

Helium: **1s²**

Lithium: **1s²2s²2p⁶3s¹**

5. Example: What does it tell us about beryllium, boron, carbon, nitrogen, and oxygen? (What does it tell us about your work, carbon? Its configuration is **1s²2s²2p²**.)

What does it tell us about boron? **1s²2s²2p¹**

6. Example: What does it tell us about the configuration of the elements sodium, magnesium, aluminum, silicon, phosphorus, sulfur, chlorine, and argon? (What does it tell us about your work, sulfur? Its configuration is **1s²2s²2p⁶3s²3p⁴**.)

Is the configuration correct? **1s²2s²2p⁶3s²3p⁴**

Show the correct configuration for the elements at right.



Gizmo Small Atoms Worksheet

Karin Nielsen-Saines

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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