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Chemical Changes

Directions: Follow the instructions to go through the simulation. Respond to the questions and prompts in the orange boxes.

Vocabulary: acid, base, catalyst, chemical change, coefficient, conservation of matter, decomposition, dissolve, double replacement, endothermic, exothermic, indicator, ion, physical change, product, reactant, single replacement, subscript, synthesis

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)



1. A student mixes baking soda and vinegar in a glass. The results are shown at left. Do you think any new substances are being created in this mixture? If so, how do you know?

I think carbon dioxide is the material that emerged from the mixture. This is due to the fact that when vinegar and baking soda are combined, a gas is produced (carbon dioxide)

2. Suppose this was done on top of a balance. Do you think the mass would change as the reaction proceeded?

If the gas is allowed to escape, the mass will decrease rather than remain constant.

3. What do you think would happen to the mass if the reaction took place inside a sealed plastic bag?

Due to its containment, the mass would remain unchanged.

Gizmo Warm-up

A **chemical change**, (or chemical reaction) occurs when one or more substances, called **reactants**, are transformed into different substances, or **products**. In the *Chemical Changes* Gizmo, you will look for evidence of chemical changes by looking at changes you can see, touch, or smell.

To begin, check that **Reactant 1** is **Sodium** and **Reactant 2** is **Water**. Sodium is a metal so soft you can cut it with a knife.



1. Click **Play** (▶). What do you observe?

The glass of sodium was removed, and the water was added. The sodium dispersed in the water, increasing the mass.

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RD Boyd

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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Table of Contents Gizmo Answer Keys Change Matter

1. Understanding the eBook Gizmo Answer Keys Change Matter
 - The Rise of Digital Reading Gizmo Answer Keys Change Matter
 - Advantages of eBooks Over Traditional Books
2. Identifying Gizmo Answer Keys Change Matter
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Gizmo Answer Keys Change Matter
 - User-Friendly Interface
4. Exploring eBook Recommendations from Gizmo Answer Keys Change Matter
 - Personalized Recommendations
 - Gizmo Answer Keys Change Matter User Reviews and Ratings
 - Gizmo Answer Keys Change Matter and Bestseller Lists
5. Accessing Gizmo Answer Keys Change Matter Free and Paid eBooks
 - Gizmo Answer Keys Change Matter Public Domain eBooks
 - Gizmo Answer Keys Change Matter eBook Subscription Services

- Gizmo Answer Keys Change Matter Budget-Friendly Options
- 6. Navigating Gizmo Answer Keys Change Matter eBook Formats
 - ePub, PDF, MOBI, and More
 - Gizmo Answer Keys Change Matter Compatibility with Devices
 - Gizmo Answer Keys Change Matter Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Gizmo Answer Keys Change Matter
 - Highlighting and Note-Taking Gizmo Answer Keys Change Matter
 - Interactive Elements Gizmo Answer Keys Change Matter
- 8. Staying Engaged with Gizmo Answer Keys Change Matter
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Gizmo Answer Keys Change Matter
- 9. Balancing eBooks and Physical Books Gizmo Answer Keys Change Matter
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Gizmo Answer Keys Change Matter
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Gizmo Answer Keys Change Matter
 - Setting Reading Goals Gizmo Answer Keys Change Matter
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Gizmo Answer Keys Change Matter
 - Fact-Checking eBook Content of Gizmo Answer Keys Change Matter
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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