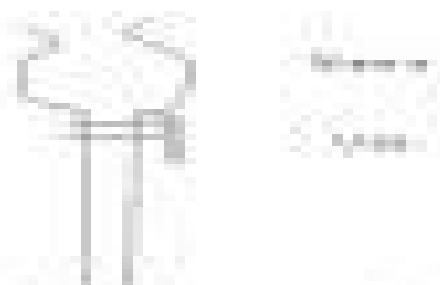


Figure 1: Chemistry: Electrode Potentials and Diagrams

Question 1:

(a) The following are standard conventional half-cells. Label them accordingly (4pts)



(b) What are measures to be taken to ensure accuracy and results of pH?

Answer 1:

There is a standard solution of three solutions: acetic acid, ammonium acetate and 4-methylpyridine.

(a) Standardized appropriate method of measurement and correct calibration is used for electrode measurement.

(b) There is a reason for the choice of your solvent.

(c) Standardization should use the highest temperature (25°C) which is appropriate for the highest

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