

PART - I: CHEMISTRY

PAPER - II

SECTION - I

Straight Objective Type

This section contains 4 multiple choice questions. Each question has 4 choices (A), (B), (C) and (D), out of which ONLY ONE is correct.

(*) marked questions are from 11th syllabus.

1. For a first order reaction $A \rightarrow P$, the temperature (T) dependent rate constant (k) was found to follow the equation $\log k = -(2000) \frac{1}{T} + 6.0$. The pre-exponential factor A and the activation energy E_a respectively,

are

(A) $1.0 \times 10^6 \text{ s}^{-1}$ and 9.2 kJ mol^{-1}

(B) 6.0 s^{-1} and 16.6 kJ mol^{-1}

(C) $1.0 \times 10^6 \text{ s}^{-1}$ and 16.6 kJ mol^{-1}

(D) $1.0 \times 10^6 \text{ s}^{-1}$ and 38.3 kJ mol^{-1}

Key.

Sol.

$$k = Ae^{-E_a/RT}$$

$$\log k = \log A - \frac{E_a}{2.303RT}$$

$$\log A = 6, A = 10^6 \text{ s}^{-1}$$

$$-\frac{E_a}{2.303 \times 8.3 \times T} = -\frac{2000}{T}$$

$$E_a = 2000 \times 2.303 \times 8.3 \text{ J}$$

$$= 38.3 \text{ kJ}$$

2. The spin only magnetic moment value (in Bohr magneton units) of $\text{Cr}(\text{CO})_6$ is

(A) 0

(B) 2.84

(C) 4.90

(D) 5.92

Key.

Sol.

(A)



CO is a strong field ligand



No. of unpaired electron = 0

\therefore magnetic moment = 0

- *3. In the following carbocation, H/CH_3 that is most likely to migrate to the positively charged carbon is



Key.

Sol.



Hydride shift from C-2 will yield resonance stabilized 2°-carbocation giving thereby ketonic product after deprotonation.

- *4. The correct stability order of the following resonance structures is



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Chandresh Agrawal,Nandini Books



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