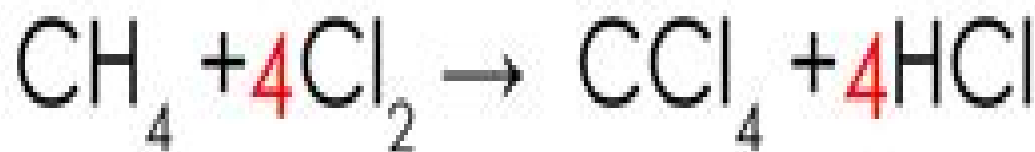


Balancing Chemical Equations Practice

Step 1 – Take Inventory of the elements and atoms on the product and reactant side.

Step 2 – Is it balanced?

Step 3 – If unbalanced, change coefficients until it's balanced.



$$\text{C} - 1 = \text{C} - 1$$

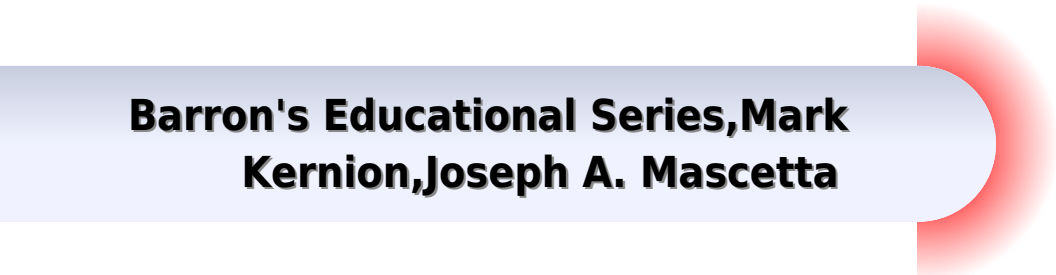
$$\text{H} - 4 \neq \text{H} - \cancel{1} 4$$

$$\text{Cl} - \cancel{2} 8 \neq \text{Cl} - \cancel{1} 8$$

Balanced!

How To Balance Chemical Equations Study Guide

**Barron's Educational Series, Mark
Kernion, Joseph A. Mascetta**



How To Balance Chemical Equations Study Guide:

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