

# ENTHALPY CHANGE STOICHIOMETRY $\Delta H$ of CHEMICAL REACTIONS

1) In the following reaction, what mass of propane must be burned to release 578.52 kJ of heat?  
 $C_3H_8(g) + 5O_2(g) \rightarrow 3CO_2(g) + 4H_2O(g) \quad \Delta H = -2,219.2 \text{ kJ}$

2) What mass of Glucose can be produced in photosynthesis if 32763 kJ of energy are consumed in the following reaction?  
 $6CO_2(g) + 6H_2O(g) \rightarrow C_6H_{12}O_6(s) + 6O_2(g) \quad \Delta H = 2838 \text{ kJ}$

3) What mass of potassium chloride can be produced when 23,834 kJ of energy are released in the following reaction:  
 $C_4H_{10}(g) + 8HClO_4(aq) \rightarrow 4KClO_4(g) + 10H_2O(l) + 8HCl(aq) \quad \Delta H = -18906 \text{ kJ}$

Name \_\_\_\_\_ Section \_\_\_\_\_ Date \_\_\_\_\_

### Enthalpy Change Stoichiometry

Directions: Solve each problem for the indicated quantity.

1) How much energy is absorbed when 6.827 grams of nitrogen is consumed in the following reaction?  
 $N_2(g) + 2O_2(g) \rightarrow 2NO_2(g) \quad \Delta H = 66.8 \text{ kJ}$

2) How much energy is released in the formation of 25.514 grams of  $Cl_2$ , using the following reaction?  
 $Cl(g) + Cl(g) \rightarrow Cl_2(g) \quad \Delta H = -98.5 \text{ kJ}$

3) Calculate the amount of energy released when acetylene burns in oxygen if 5.62 grams of acetylene are burned in the following reaction:  
 $2C_2H_2(g) + 5O_2(g) \rightarrow 4CO_2(g) + 2H_2O(g) \quad \Delta H = -2599.2 \text{ kJ}$

4) How much energy is released when 13.82 grams of isooctane reacts with oxygen? Use the following reaction:  
 $2C_8H_{18}(l) + 25O_2(g) \rightarrow 16CO_2(g) + 18H_2O(g) \quad \Delta H = -10912 \text{ kJ}$

5) How much energy is released when 2090 grams of  $NO_2$  are formed in the following reaction:  
 $2NO(g) + 2HCl(aq) \rightarrow 2NOCl(aq) + 2H_2(g) \quad \Delta H = -3049 \text{ kJ}$

6) What mass of titanium(IV) chloride is consumed in the following reaction if 583.47 kJ of energy?  
 $TiCl_4(g) + 2H_2O(g) \rightarrow TiO_2(s) + 4HCl(g) \quad \Delta H = 87.5 \text{ kJ}$

7) What mass of Chlorine gas is consumed in the following reaction if 397.8 kJ are released?  
 $Cl_2(g) + 3Cl(g) \rightarrow CCl_4(g) + 3Cl_2(g) \quad \Delta H = -130.3 \text{ kJ}$

## A CHEMISTRY WORKSHEET

## 10 PROBLEMS

## 3 VERSIONS

## WORKED ANSWER KEY

# Heat Of Reaction Answer Key

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Heat Of Reaction Answer Key: Bestsellers in 2023 The year 2023 has witnessed a noteworthy surge in literary brilliance, with numerous compelling novels enthralling the hearts of readers worldwide. Lets delve into the realm of bestselling books, exploring the captivating narratives that have enthralled audiences this year. Heat Of Reaction Answer Key : Colleen Hoover's "It Ends with Us" This heartfelt tale of love, loss, and resilience has captivated readers with its raw and emotional exploration of domestic abuse. Hoover masterfully weaves a story of hope and healing, reminding us that even in the darkest of times, the human spirit can succeed. Uncover the Best : Taylor Jenkins Reids "The Seven Husbands of Evelyn Hugo" This captivating historical fiction novel unravels the life of Evelyn Hugo, a Hollywood icon who defies expectations and societal norms to pursue her dreams. Reids absorbing storytelling and compelling characters transport readers to a bygone era, immersing them in a world of glamour, ambition, and self-discovery. Discover the Magic : Delia Owens "Where the Crawdads Sing" This captivating coming-of-age story follows Kya Clark, a young woman who grows up alone in the marshes of North Carolina. Owens spins a tale of resilience, survival, and the transformative power of nature, entrancing readers with its evocative prose and mesmerizing setting. These top-selling novels represent just a fraction of the literary treasures that have emerged in 2023. Whether you seek tales of romance, adventure, or personal growth, the world of literature offers an abundance of engaging stories waiting to be discovered. The novel begins with Richard Papen, a bright but troubled young man, arriving at Hampden College. Richard is immediately drawn to the group of students who call themselves the Classics Club. The club is led by Henry Winter, a brilliant and charismatic young man. Henry is obsessed with Greek mythology and philosophy, and he quickly draws Richard into his world. The other members of the Classics Club are equally as fascinating. Bunny Corcoran is a wealthy and spoiled young man who is always looking for a good time. Charles Tavis is a quiet and reserved young man who is deeply in love with Henry. Camilla Macaulay is a beautiful and intelligent young woman who is drawn to the power and danger of the Classics Club. The students are all deeply in love with Morrow, and they are willing to do anything to please him. Morrow is a complex and mysterious figure, and he seems to be manipulating the students for his own purposes. As the students become more involved with Morrow, they begin to commit increasingly dangerous acts. The Secret History is a brilliant and suspenseful novel that will keep you speculating until the very end. The novel is a cautionary tale about the dangers of obsession and the power of evil.

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