

## Heat Capacity Practice Problems-Worksheet 4

Name: \_\_\_\_\_ Period: \_\_\_\_\_

SHOW ALL WORK FOR FULL CREDIT! Remember to include units in your answers. Round answers to the proper number of significant figures.

Substance	Specific Heat (J/g x °C)
Water	4.18
Aluminum	0.90
Iron	0.46
Steam	1.7
Mercury	0.14
Chloroform	0.96
Grain Alcohol	2.4
Silver	0.24
Ice	2.1

1. How much heat will be absorbed by 25g of chloroform when its temperature started at 67.0°C and after it was 112.0°C?

Calculator answer: \_\_\_\_\_

SF answer: \_\_\_\_\_

2. How much heat will be absorbed by 130g of Mercury when its temperature started at 24.0°C and after it was 78.0°C?

Calculator answer: \_\_\_\_\_

SF answer: \_\_\_\_\_

3. If the temperature of a 95.00 g sample of Iron started at 36.0°C and ended at 58.0°C, how much heat would be lost?

Calculator answer: \_\_\_\_\_

SF answer: \_\_\_\_\_

4. What is the definition of Temperature? \_\_\_\_\_

\_\_\_\_\_

5. What is the definition of Heat? \_\_\_\_\_

\_\_\_\_\_

6. Why can there never be a temperature less than 0 Kelvins? \_\_\_\_\_

\_\_\_\_\_

# Heat Capacity Practice Problems 4 Answers

**James Hasik, Stacey Rudnick, Ryan  
Hackney**



## Heat Capacity Practice Problems 4 Answers:

**Survival Guide to General Chemistry** Patrick E. McMahon, Rosemary McMahon, Bohdan Khomtchouk, 2019-02-13 This work evolved over thirty combined years of teaching general chemistry to a variety of student demographics. The focus is not to recap or review the theoretical concepts well described in the available texts. Instead, the topics and descriptions in this book make available specific detailed step by step methods and procedures for solving the major types of problems in general chemistry. Explanations, instructional process sequences, solved examples, and completely solved practice problems are greatly expanded, containing significantly more detail than can usually be devoted to in a comprehensive text. Many chapters also provide alternative viewpoints as an aid to understanding. Key Features: The authors have included every major topic in the first semester of general chemistry and most major topics from the second semester. Each is written in a specific and detailed step by step process for problem solving, whether mathematical or conceptual. Each topic has greatly expanded examples and solved practice problems containing significantly more detail than found in comprehensive texts. Includes a chapter designed to eliminate confusion concerning acid base reactions, which often persists through working with acid base equilibrium. Many chapters provide alternative viewpoints as an aid to understanding. This book addresses a very real need for a large number of incoming freshmen in STEM fields. *Physics Daily Practice Problem Book* Career Point Kota, 2019-11-18 E3 Chemistry Guided Study Book - 2018 Home Edition (Answer Key Included) Effiong Eyo, 2017-12-08 Chemistry students and Homeschoolers Go beyond just passing. Enhance your understanding of chemistry and get higher marks on homework, quizzes, tests, and the regents exam with E3 Chemistry Guided Study Book 2018. With E3 Chemistry Guided Study Book, students will get clean, clear, engaging, exciting, and easy to understand high school chemistry concepts with emphasis on New York State Regents Chemistry. The Physical Setting Easy to read format to help students easily remember key and must know chemistry materials. Several example problems with guided step by step solutions to study and follow. Practice multiple choice and short answer questions along side each concept to immediately test student understanding of the concept. 12 topics of Regents question sets and 2 most recent Regents exams to practice and prep for any Regents Exam. This is the Home Edition of the book. Also available in School Edition. ISBN 978 1979088374. The Home Edition contains answer key to all questions in the book. Teachers who want to recommend our Guided Study Book to their students should recommend the Home Edition. Students and parents whose school is not using the Guided Study Book as instructional material, as well as homeschoolers, should also buy the Home edition. The School Edition does not have the answer key in the book. A separate answer key booklet is provided to teachers with a class order of the book. Whether you are using the school or Home Edition, our E3 Chemistry Guided Study Book makes a great supplemental instructional and test prep resource that can be used from the beginning to the end of the school year. PLEASE NOTE: Although reading contents in both the school and home editions are identical, there are slight differences in question numbers, choices, and pages between the two editions. Students whose

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An Inquiry-Based Introduction to Engineering Michelle Blum, 2022-09-20 The text introduces engineering to first year undergraduate students using Inquiry Based Learning (IBL). It draws on several different inquiry based instruction types such as confirmation inquiry, structured inquiry, guided inquiry and open inquiry and all of their common elements. Professor Blum's approach emphasizes the student's role in the learning process, empowering them in the classroom to explore the material, ask questions and share ideas instead of the instructor lecturing to passive learners about what they need to know. Beginning with a preface to IBL, the book is organized into three parts, each consisting of four to ten chapters. Each chapter has a dedicated topic where an initial few paragraphs of introductory or fundamental material are provided. This is followed by a series of focused questions

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