

Guided Reading (Chapter 18 Section 2)

1. What do chemical bonds do?

Chemical bonds hold the atoms of molecules together and determine their chemical and physical properties.

2. Is a covalent bond more often between atoms that are nonmetals or metals? Explain.

3. Why do some molecules accept H-bonds while others do not? Give an example of a molecule that does and one that does not.

The water molecules are hydrogen-bonded with its structure being its shape and its H₂O.

4. Which group of elements is called the noble gases? Group 18

5. Is water ionic or covalent? Explain why you think so.

6. What is a hydrogen bond?

The bonds result when one of the atoms of one molecule is attracted to another molecule.

7. Why are the water gas molecules attracted to each other?

Because they are continuously moving and interacting.

8. How many electrons does the hydrogen atom have? How many electrons does it need to be stable?

One electron

9. What are the high-energy bond electrons of an atom called?

Valence electrons

Guided Reading Chapter 16 Section Covalent

Ian Pickup



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