

ACS Final Exam Practice 1

The exam is 70 questions in 110 minutes. You may use the equation sheet and periodic table but no other materials

Section 1 Foundations

PQ-1. The speed of light is $3.0 \times 10^8 \text{ m}\cdot\text{s}^{-1}$. What is this speed in $\text{nm}\cdot\text{ms}^{-1}$?

- (A) $3.0 \times 10^{-1} \text{ nm}\cdot\text{ms}^{-1}$ (B) $3.0 \times 10^2 \text{ nm}\cdot\text{ms}^{-1}$ (C) $3.0 \times 10^{14} \text{ nm}\cdot\text{ms}^{-1}$ (D) $3.0 \times 10^{23} \text{ nm}\cdot\text{ms}^{-1}$

PQ-4. What is the correctly reported mass of water based on the data in the table?

Mass of beaker and water	29.62 g
Mass of beaker only	28.3220 g

- (A) 1.3 g (B) 1.30 g (C) 1.298 g (D) 1.2980 g

PQ-5. What is the name of $\text{Ti}_2(\text{PO}_4)_3$?

- (A) titanium phosphate (B) titanium(III) phosphate
(C) titanium(IV) phosphate (D) titanium tetraphosphate

PQ-6. The formula of strontium hexafluorosilicate is SrSiF_6 . What is the formula of aluminum hexafluorosilicate?

- (A) AlSiF_6 (B) $\text{Al}_2(\text{SiF}_6)_3$ (C) Al_2SiF_6 (D) $\text{Al}_2(\text{SiF}_6)_2$

PQ-8. What is the correct name for N_2O_5 ?

- (A) nitrogen(II) oxide (B) nitrogen(V) oxide
(C) dinitrogen oxide (D) dinitrogen pentoxide

PQ-12. If a palladium nanoparticle has a density of $12.0 \text{ g}\cdot\text{cm}^{-3}$, what is the mass of a nanoparticle with a volume of $1.84 \times 10^{-23} \text{ cm}^3$?

- (A) $2.21 \times 10^{-23} \text{ g}$ (B) $1.63 \times 10^{-20} \text{ g}$ (C) $6.52 \times 10^{-21} \text{ g}$ (D) $1.53 \times 10^{-22} \text{ g}$

Guided Practice Problem Chemistry 1page 360 Answers

Peipei Pang



Guided Practice Problem Chemistry 1page 360 Answers:

Guided Practice Problem Chemistry 1page 360 Answers Book Review: Unveiling the Magic of Language

In an electronic era where connections and knowledge reign supreme, the enchanting power of language has been apparent than ever. Its capability to stir emotions, provoke thought, and instigate transformation is truly remarkable. This extraordinary book, aptly titled "**Guided Practice Problem Chemistry 1page 360 Answers**," written by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we shall delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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Table of Contents Guided Practice Problem Chemistry 1page 360 Answers

1. Understanding the eBook Guided Practice Problem Chemistry 1page 360 Answers
 - The Rise of Digital Reading Guided Practice Problem Chemistry 1page 360 Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Guided Practice Problem Chemistry 1page 360 Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Guided Practice Problem Chemistry 1page 360 Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Guided Practice Problem Chemistry 1page 360 Answers
 - Personalized Recommendations
 - Guided Practice Problem Chemistry 1page 360 Answers User Reviews and Ratings
 - Guided Practice Problem Chemistry 1page 360 Answers and Bestseller Lists

5. Accessing Guided Practice Problem Chemistry 1page 360 Answers Free and Paid eBooks
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 - Guided Practice Problem Chemistry 1page 360 Answers eBook Subscription Services
 - Guided Practice Problem Chemistry 1page 360 Answers Budget-Friendly Options
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 - ePub, PDF, MOBI, and More
 - Guided Practice Problem Chemistry 1page 360 Answers Compatibility with Devices
 - Guided Practice Problem Chemistry 1page 360 Answers Enhanced eBook Features
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 - Adjustable Fonts and Text Sizes of Guided Practice Problem Chemistry 1page 360 Answers
 - Highlighting and Note-Taking Guided Practice Problem Chemistry 1page 360 Answers
 - Interactive Elements Guided Practice Problem Chemistry 1page 360 Answers
8. Staying Engaged with Guided Practice Problem Chemistry 1page 360 Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Guided Practice Problem Chemistry 1page 360 Answers
9. Balancing eBooks and Physical Books Guided Practice Problem Chemistry 1page 360 Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Guided Practice Problem Chemistry 1page 360 Answers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Guided Practice Problem Chemistry 1page 360 Answers
 - Setting Reading Goals Guided Practice Problem Chemistry 1page 360 Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Guided Practice Problem Chemistry 1page 360 Answers
 - Fact-Checking eBook Content of Guided Practice Problem Chemistry 1page 360 Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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