

**CHAPTER 6 REVIEW***Chemical Bonding***MIXED REVIEW**

**SHORT ANSWER** Answer the following questions in the space provided.

1. Name the type of energy that is a measure of strength for each of the following types of bonds:

_____ <b>lattice energy</b> _____	<b>a.</b> ionic bond
_____ <b>bond energy</b> _____	<b>b.</b> covalent bond
_____ <b>enthalpy of vaporization</b> _____	<b>c.</b> metallic bond

2. Use the electronegativity values shown in **Figure 20**, on page 161 of the text, to determine whether each of the following bonds is nonpolar covalent, polar covalent, or ionic.

_____ <b>ionic</b> _____	<b>a.</b> H—F	_____ <b>nonpolar covalent</b> _____	<b>d.</b> H—H
_____ <b>ionic</b> _____	<b>b.</b> Na—Cl	_____ <b>polar covalent</b> _____	<b>e.</b> H—C
_____ <b>polar covalent</b> _____	<b>c.</b> H—O	_____ <b>polar covalent</b> _____	<b>f.</b> H—N

3. How is a hydrogen bond different from an ionic or covalent bond?

A hydrogen bond is a dipole-dipole attraction between a partially positive hydrogen  
atom and the unshared electron pair of a strongly electronegative atom such as O,  
N, or F. Unlike ionic or covalent bonds, in which electrons are given up or shared,  
the hydrogen bond is a weaker attraction. Hydrogen bonds are generally  
intermolecular, while ionic and covalent bonds occur between ions or atoms  
respectively.

4. H<sub>2</sub>S and H<sub>2</sub>O have similar structures and their central atoms belong to the same group. Yet H<sub>2</sub>S is a gas at room temperature and H<sub>2</sub>O is a liquid. Use bonding principles to explain why this is.

Oxygen has higher electronegativity than sulfur, which creates a highly polar bond.  
Increased polarity in H<sub>2</sub>O bonds means a stronger intermolecular attraction, making  
water a liquid at room temperature. Hydrogen bonding exists between water  
molecules, but not between hydrogen sulfide molecules.

# Guided And Study Workbook Answers Covalent Bonding

**RS Peters**



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**Study Guide for The Human Body in Health & Disease** Linda Swisher, RN, EdD, Kevin T. Patton, PhD, Gary A. Thibodeau, PhD, 2013-02-15 Reinforce your understanding of the concepts in Patton and Thibodeau s The Human Body in Health Disease 6th Edition Corresponding to the chapters in the text this study guide reviews essential medical terminology concepts and processes related to the anatomy and physiology of the human body and body function in health and disease A variety of exercises make it easy to review and apply key concepts and labeling of anatomy drawings helps you learn anatomical structures and terminology UPDATED Did You Know provides fun interesting facts on A P topics A brief synopsis at the beginning of each chapter previews core concepts that will be covered Crossword Puzzle Unscramble and Word Find activities help you learn new vocabulary terms and their proper spelling Diagrams and labeling exercises

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**Study Guide for The Human Body in Health & Disease - E-Book** Kevin T. Patton, Frank B. Bell, Terry Thompson, Peggie L. Williamson, Eric L. Sun, 2023-03-02 Reinforce your understanding of the concepts in Patton's The Human Body in Health Disease Corresponding to the chapters in the text this study guide reviews essential medical terminology concepts and processes related to anatomy and physiology and explains how body systems function in health and disease Each chapter begins with a quick synopsis of the key points in the textbook chapter A variety of exercises make it easier to review and apply key concepts and labeling of anatomy drawings helps in learning anatomical terms and structures Know your Medical Terms feature helps familiarize you with the various word parts used in medical terminology and pairs nicely with the Language of Medicine word lists in the main text to reinforce medical terminology concepts that are key to understanding A P Brief synopsis of core textbook concepts provides a comprehensive review of essential content Crossword Puzzles and Word Finds help you master new vocabulary terms Application Questions ask you to make judgments based on the information in the chapter Diagrams and labeling exercises help reinforce where the structures of the body are located Matching and fill in the blank exercises aid in understanding chapter content Answers to exercises are provided in the back of the guide along with references to the appropriate textbook page and the textbook objective so that you can easily find where the material is presented Study tips in the Preface provide an overview of the most effective methods for learning and retaining information

**Organic Chemistry Study Guide** Robert J. Ouellette, J. David Rawn, 2014-11-04 Organic Chemistry Study Guide Key Concepts Problems and Solutions features hundreds of problems from the companion book Organic Chemistry and includes solutions for every problem Key concept summaries reinforce critical material from the primary book and enhance mastery of this complex subject Organic chemistry is a constantly evolving field that has great relevance for all scientists not just chemists For chemical engineers understanding the properties of organic molecules and how reactions occur is critically important to understanding the processes in an industrial plant For biologists and health professionals it is essential because nearly all of biochemistry springs from organic chemistry Additionally all scientists can

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**Hazmat Chemistry Study Guide (Second Edition)** Jill Meryl Levy, 2011

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