

# Lab 11

## Gravimetric Analysis

### Pre-Lab Questions

1. Write a balanced equation for the precipitation of calcium carbonate from potassium carbonate and calcium chloride.  
$$\text{K}_2\text{CO}_3 + \text{CaCl}_2 \rightarrow \text{CaCO}_3 + 2\text{KCl}$$
2. Using this balanced equation, determine the limiting reactant if 15 grams of calcium chloride was reacted with 15 grams of potassium carbonate.

Using your answer for Question 2, determine the mass of potassium carbonate needed to fully precipitate all the calcium from a 25 mL sample of 15% calcium chloride.

# Gravimetric Analysis Prelab Answers

**László Erdey, Ilona Buzás, Gyula Svehla**

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Working with Chemistry Donald J. Wink, Sharon Fetzer-Gislason, Julie Ellefson Kuehn, 2004-02-20 With this modular laboratory program students build skills using important chemical concepts and techniques to the point where they are able to design a solution to a scenario drawn from a professional environment The scenarios are drawn from the lives of people who work with chemistry every day ranging from field ecologists to chemical engineers and include many health professionals as well     **Instructors Manual to Lab Manual** Ralph Petrucci, William Harwood, Geoffrey Herring, 2001  
Academic Computing , 1987     **Investigation in General Chemistry** Alice S. Cohen, 1978     **Book of Abstracts** American Chemical Society. Meeting, American Chemical Society, 1974     **Laboratory Experiments for General Chemistry** Harold R. Hunt, Toby F. Block, 1994     **General Chemistry** Hunt, 1976-06-07     **Gravimetric Analysis** László Erdey, 1963     **Gravimetric Analysis** László Erdey, 2013-10-22 Gravimetric Analysis Part III describes the experimental procedures for the gravimetric analysis of various compounds This book is composed of 13 chapters that also present sample preparation protocols The first four chapters survey the steps for halogen compound determination The succeeding chapters provide the procedures for gravimetric determination of cyanide thiocyanate ions sulfur nitrogen phosphorus carbon silicon and boron The final chapter considers other aspects of gravimetric experiments including apparatus cleaning reagents and numerical calculation of the result This book will prove useful to analytical and inorganic chemists teachers and students in the allied fields     Gravimetric Analysis László Erdey, 2013-09-03 Analytical Chemistry Volume 7 Gravimetric Analysis Part II describes the experimental procedures for the gravimetric analysis of Groups I to V cations This book is composed of 43 chapters that also present sample preparation separation and precipitation protocols The first six chapters include Group I cations such as silver lead mercury copper bismuth and cadmium followed by chapters on Group II cations including arsenic antimony tin germanium gold platinum selenium and tellurium The subsequent chapters explore the gravimetric determination of Group III cations namely aluminum iron chromium nickel cobalt zinc manganese titanium zirconium hafnium thorium scandium niobium and tantalum molybdenum tungsten vanadium uranium thallium indium gallium and beryllium The remaining chapters are devoted to analysis of various forms of Groups IV and V cations This book will prove useful to analytical and inorganic chemists teachers and students in the allied fields     **Introductory Titrimetric and Gravimetric Analysis** Evelyn M. Rattenbury, 2016-06-06 Introductory Titrimetric and Gravimetric Analysis discusses the different types of titration and the weighing of different solutions in solid form Coverage is made on acid base titration argentometric titrations and oxidation reduction titrations Iodometric titrations and complexometric titrations are also explained Extensive discussion on each of the titration method along with some examples and laboratory experiments is given The process of weight measurement of damp powder is one example of the experiments The book is a manual that guides a student to the correct ways of conducting an experiment made on such solutions as sodium hydroxide using

hydrochloric acid and oxalic acid Outcome of such experiments in terms of composition weight of solutions and measurement of pressure in certain environment is tabulated and briefly explained Logarithms and antilogarithms are included at the end of the book The text will serve as a good laboratory manual for students preparing for science examination as well as for chemists and chemical engineers     **Gravimetric Analysis** Laszlo Erdey,1963     Gravimetric analysis László Erdey,Ilona Buzás,Gyula Svehla,1963     *Gravimetric Analysis; 1* László Erdey,2021-09-09 This work has been selected by scholars as being culturally important and is part of the knowledge base of civilization as we know it This work is in the public domain in the United States of America and possibly other nations Within the United States you may freely copy and distribute this work as no entity individual or corporate has a copyright on the body of the work Scholars believe and we concur that this work is important enough to be preserved reproduced and made generally available to the public To ensure a quality reading experience this work has been proofread and republished using a format that seamlessly blends the original graphical elements with text in an easy to read typeface We appreciate your support of the preservation process and thank you for being an important part of keeping this knowledge alive and relevant     **Introductory Titrimetric and Gravimetric Analysis** Evelyn M. Rattenbury,2013 Introductory Titrimetric and Gravimetric Analysis is an introductory course to the general principles of titrimetric and gravimetric analysis of chemical solutions and substances The book is divided in two parts Part I focuses on the principles of titrimetric analysis covering topics on acid base argentometric oxidation reduction iodometric and complexometric titrations Part II covers the principles of gravimetric analysis discussing gravimetric process and methods and the gravimetric determinations of several chemical solutions and substances Chemists and students of chemistry wil     **Supplementary Notes on Gravimetric Analysis for Beginners** W. Lowson,1923     **Gravimetric Analysis. Pt. 2** L. Erdey,1965     **Gravimetric Analysis** Lásxlé Erdey,1963     **Gravimetric analysis. 2** László Erdey,1965  
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