

Chapter Assessment Chemistry: Matter and Change • Chapter 4

Reviewing Vocabulary

Match each definition in Column A with the term in Column B.

Column A

Column B

- Radiation deflected toward the positively charged plate.
 - Atoms with the same number of protons but different numbers of neutrons.
 - High-energy radiation that has no charge and no mass.
 - The smallest particle of an element that retains the properties of that element.
 - The weighted average mass of an element's isotopes.
 - The center-most part of an atom where the protons and neutrons are contained.
 - Radiation deflected toward the negatively charged plate.
 - The rays and particles emitted by radioactive material.
 - Equal to $1/12$ the mass of a carbon-12 atom.
 - Ray of radiation traveling from the cathode to the anode.
 - Process not requiring energy by which unstable nuclei lose energy.
 - States that all matter is composed of atoms.
 - Process whereby some substances spontaneously emit radiation.
- Compare and contrast each pair of related terms.
- mass number, atomic number
 - nuclear reaction, nuclear equation

- atom
- nucleus
- atomic mass
- isotope
- gamma ray
- alpha radiation
- beta
- atomic
- isotope
- Dalton's atomic theory
- cathode ray
- isotope
- radioactivity

Understanding Main Ideas

Use the periodic table to identify each element described below.

- atomic number 63
- 18 protons
- 44 protons and 44 electrons
- atomic number 24
- 21 protons
- atomic number 55

In the space at the left, write true if the statement is true; if the statement is false,

change the italicized term to make it true.

- An atom's nucleus contains its protons and electrons.
- Neutrons have no electrical charge.
- Beta particles have a charge of 2^+ .
- An alpha particle consists of two protons and two electrons.

Complete the table below.

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Solutions Manual for Chemistry: Molecules Matter and Change, Fourth Edition Julie Henderleiter, Charles Trapp, Peter Atkins, Loretta Jones, 1999-08-13 This student companion is a supplement to Chemistry Molecules Matter and Change 4th edition with CD ROM It features guided reading strategies collaborative learning sheets and strategies for using CD ROM tools

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