Name: Amanda Brito Date: 04/29/2019

Student Exploration: Equilibrium and Concentration

Vocabulary: chemical equilibrium, concentration, equilibrium, equilibrium constant, reaction quotient, reversible reaction

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)
Gary has \$5,000 in his bank account and earns a modest salary. Every month he pays for rent, food, utilities, and entertainment.

- A. How will Gary's account change if he saves more than he spends? _it will increase.
- B. How will Gary's account change if he spends more than he saves? It will decrease.
- C. What happens if Gary spends exactly as much as he saves? It will stay the same.

Gizmo Warm-up

If Gary spends exactly as much as he earns, his savings will be in **equilibrium**. Equilibrium occurs when two opposing processes occur at the same rate, leading to no net change. In the Equilibrium and Concentration Gizmo™, you will investigate how equilibrium can occur in chemical reactions.

To begin, check that Reaction 1 is selected. Set Moles NO_2 to 8 and Moles N_2O_4 to 0.



Click Play () and observe the colliding molecules. What do you notice?
 the molecules move and collide. Occasionally there is a blue or red flash.

In the Gizmo, a blue flash appears every time two reactants combine to form a product. A red flash appears every time a product dissociates into reactants.

Click Reset (□), and set Moles NO₂ to 0 and Moles N₂O₄ to 8. Click Play.

What do you notice now? There are more red flashes than blue.

When a reaction can proceed in either direction, it is a reversible reaction. Based on what
you have observed, is the synthesis of NO₂ into N₂O₄ a reversible reaction? Explain.

I know the synthesis of NO2 into N2O4 is a reversible reaction because both the forward reaction (blue flashes) and the reverse reaction (red flashes) are observed.



Gizmo Equilibrium And Concentration Answers

Padhraic Smyth

Gizmo Equilibrium And Concentration Answers:

An Introduction to Equilibrium Problems ,2016 You use your basic understanding of equilibrium systems to try to solve some problems You tackle two kinds of equilibrium problems ones in which you are asked to calculate the equilibrium constant for an equation and ones in which you are asked to find the equilibrium concentration of a reactant or product

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- 1. Understanding the eBook Gizmo Equilibrium And Concentration Answers
 - The Rise of Digital Reading Gizmo Equilibrium And Concentration Answers
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Gizmo Equilibrium And Concentration Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Gizmo Equilibrium And Concentration Answers
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Gizmo Equilibrium And Concentration Answers
 - Personalized Recommendations
 - Gizmo Equilibrium And Concentration Answers User Reviews and Ratings

- Gizmo Equilibrium And Concentration Answers and Bestseller Lists
- 5. Accessing Gizmo Equilibrium And Concentration Answers Free and Paid eBooks
 - Gizmo Equilibrium And Concentration Answers Public Domain eBooks
 - Gizmo Equilibrium And Concentration Answers eBook Subscription Services
 - Gizmo Equilibrium And Concentration Answers Budget-Friendly Options
- 6. Navigating Gizmo Equilibrium And Concentration Answers eBook Formats
 - ∘ ePub, PDF, MOBI, and More
 - Gizmo Equilibrium And Concentration Answers Compatibility with Devices
 - Gizmo Equilibrium And Concentration Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Gizmo Equilibrium And Concentration Answers
 - Highlighting and Note-Taking Gizmo Equilibrium And Concentration Answers
 - Interactive Elements Gizmo Equilibrium And Concentration Answers
- 8. Staying Engaged with Gizmo Equilibrium And Concentration Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Gizmo Equilibrium And Concentration Answers
- 9. Balancing eBooks and Physical Books Gizmo Equilibrium And Concentration Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Gizmo Equilibrium And Concentration Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Gizmo Equilibrium And Concentration Answers
 - Setting Reading Goals Gizmo Equilibrium And Concentration Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Gizmo Equilibrium And Concentration Answers
 - Fact-Checking eBook Content of Gizmo Equilibrium And Concentration Answers
 - Distinguishing Credible Sources

- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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