

1. Circle the element with the largest atomic radius and put a square around the element with the smallest atomic radius:

Cu **K** Ni **Br**

a. Explain why you made these choices: *All of the elements are in the same period. The trend in atomic radius as you go across a period is DECREASING. Therefore, the element on the far left (K) is the largest, and the element on the far right (Br) is the smallest.*

2. Circle the element with the highest ionization energy and put a square around the element with the lowest ionization energy:

Cu **K** Ni **Br**

a. Explain why you made these choices: *All of the elements are in the same period. The trend in ionization energy as you go across a period is INCREASING. Therefore, the element on the far left (K) has the lowest ionization energy, and the element on the far right (Br) has the highest ionization energy.*

3. Circle the element with the highest electronegativity and put a square around the element with the lowest electronegativity:

Cu **K** Ni **Br**

a. Explain why you made these choices: *All of the elements are in the same period. The trend in electronegativity as you go across a period is INCREASING. Therefore, the element on the far left (K) has the lowest electronegativity, and the element on the far right (Br) has the highest electronegativity.*

4. For each of the following groups: Circle the element with the largest atomic radius and put a square around the element with the smallest atomic radius:

- 5.
- a. O C **Be** **Ne** Same Period
 - b. Na Rb **Fr** **H** Same Group
 - c. **Pb** **C** Sn Si Same Group
 - d. Au W S **Fr** **Ne** Zn Challenge

6. For each of the following groups: Circle the element with the highest ionization energy and put a square around the element with the lowest ionization energy:

- a. O C **Be** **Ne** Same Period
- b. Na Rb **Fr** **H** Same Group
- c. **Pb** **C** Sn Si Same Group
- d. Au W S **Fr** **Ne** Zn Challenge

Electronegativity Periodic Trends Practice Answers

Rosina Ehmman



Electronegativity Periodic Trends Practice Answers:

The book delves into Electronegativity Periodic Trends Practice Answers. Electronegativity Periodic Trends Practice Answers is a crucial topic that must be grasped by everyone, from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Electronegativity Periodic Trends Practice Answers, encompassing both the fundamentals and more intricate discussions.

1. This book is structured into several chapters, namely:
 - Chapter 1: Introduction to Electronegativity Periodic Trends Practice Answers
 - Chapter 2: Essential Elements of Electronegativity Periodic Trends Practice Answers
 - Chapter 3: Electronegativity Periodic Trends Practice Answers in Everyday Life
 - Chapter 4: Electronegativity Periodic Trends Practice Answers in Specific Contexts
 - Chapter 5: Conclusion
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