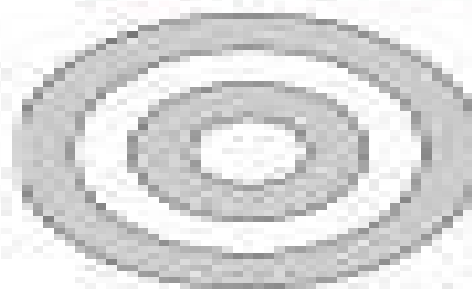


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Read: <http://www.ck12.org/Book-Search/Book/CK-12-Science-Concepts-Book/Chapter/Chapter%2016%20-%20Electron%20Configuration/>

1. Label the components of the diagram:



2. How many energy levels does hydrogen possess? _____
3. How many energy levels does chlorine possess? _____
4. Describe the Aufbau principle: _____

Read: <http://www.ck12.org/Book-Search/Book/CK-12-Science-Concepts-Book/Chapter/Chapter%2016%20-%20Electron%20Configuration/>

5. What is an orbital? _____
6. What is an s-orbital? _____
7. What is a p-orbital? _____
8. How many orbitals are there? _____
9. Describe Hund's Rule: _____

10. What are the other two levels of orbitals? _____

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11. What is the electron configuration for Fe? _____
12. Describe electron configuration notation and noble gases: _____

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Julius Sumner Miller



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