

PHYSICAL CHEMISTRY

(Complete Syllabus)

SECTION - A

Straight Objective Type

This section contains 220 multiple choice questions. Each question has 4 choices (1), (2), (3) and (4), out of which **ONLY ONE** is correct.

Choose the correct answer :

- The mass of oxygen required to produce enough CO which completely reduces x gm of Fe_2O_3 . Suppose molecular weight of $\text{O}_2 = y$, molecular weight of $\text{Fe}_2\text{O}_3 = z$
(1) $\frac{2xz}{3y}$ (2) $\frac{3xy}{2z}$
(3) $\frac{2y}{3xz}$ (4) $\frac{3z}{2xy}$
- The number of lone pair of electrons present in 2.56 gm of orthorhombic sulphur will be
(1) 0.02 N_0 (2) 0.16 N_0
(3) 0.32 N_0 (4) 0.08 N_0
- A 10.0 g of sample of mixture of calcium chloride and sodium chloride is treated with Na_2CO_3 to precipitate the calcium as calcium carbonate. This CaCO_3 is heated to convert all calcium to CaO and final mass of CaO is 1.62 gms. The % by mass of CaCl_2 in the original mixture is
(1) 15.2%
(2) 32.1%
(3) 21.8%
(4) 11.07%
- 1000 ml of oxygen at STP were passed through an ozoniser and the resulting volume was 888 ml at STP. This quantity of ozonised oxygen is passed through excess of KI solution. The weight of I_2 liberated is
(1) 25.4 g (2) 12.7 g
(3) 2.54 g (4) 1.27 g
- x_1 ml solution having normality 1 is diluted to get a solution of normality y_2 . What would be volume required for dilution?
(1) $x_1 \left(1 - \frac{y_1}{y_2}\right)$ (2) $x_1 \left(\frac{y_2}{y_1} - 1\right)$
(3) $x_1 \left(\frac{y_1}{y_2} - 1\right)$ (4) $x_1 \left(1 + \frac{y_1}{y_2}\right)$
- 1.625 g of an interhalogen compound ICl_x was subjected to electric discharge which split it into constituents and brought to STP. It occupied 112 ml. The interhalogen compound is
(1) ICl (2) ICl_3
(3) ICl_5 (4) ICl_7
- If 12.25 g of a sample of KClO_3 of 75% purity and rest vapourisable impurities is strongly heated to give KCl , KClO_4 and 0.32 gm O_2 . The mass of residue obtained is
(1) 1.769 g (2) 7.096 g
(3) 5.327 g (4) 8.85 g
- A 2 g sample of xenon reacts with fluorine. The mass of compound produced is 3.158 g. The empirical formula of the compound is
(1) XeF_2 (2) XeF_4
(3) XeF_6 (4) XeF_8
- 1 g of carbonate of metal was dissolved in 25 ml of 1 N HCl . The resulting liquid required 5 ml of 1 N NaOH for neutralisation. The equivalent weight

Multiple Choice Physical Chemistry Exam Questions

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dilution of solution solubility solutions suspension and colloids and types of solutions The Structure of Atoms MCQ PDF e Book Chapter 7 practice test to solve MCQ questions on Atomic structure experiments electronic configuration and isotopes The Structure of Molecules MCQ PDF e Book Chapter 8 practice test to solve MCQ questions on Atoms reaction bonding nature and properties chemical bonds intermolecular forces and types of bonds **Physical Chemistry** Kenneth S Schmitz, 2018-06-14 The advancements in society are intertwined with the advancements in science To understand how changes in society occurred and will continue to change one has to have a basic understanding of the laws of physics and chemistry Physical Chemistry Multidisciplinary Applications in Society examines how the laws of physics and chemistry physical chemistry explain the dynamic nature of the Universe and events on Earth and how these events affect the evolution of society multidisciplinary applications The ordering of the chapters reflects the natural flow of events in an evolving Universe Philosophy of Science the basis of the view that natural events have natural causes Cosmology the origin of everything from the Big Bang to the current state of the Universe Geoscience the physics and chemistry behind the evolution of the planet Earth from its birth to the present Life Science the molecules and mechanisms of life on Earth Ecology the interdependence of all components within the Ecosphere and the Universe Information Content emphasis on how words and phrases and framing of issues affect opinions reliability of sources and the limitations of knowledge Addresses the four Ws of science Why scientists believe Nature works the way it does Who helped develop the fields of science What theories of natural processes tell us about the nature of Nature and Where our scientific knowledge is taking us into the future Gives a historical review of the evolution of science and the accompanying changes in the philosophy of how science views the nature of the Universe Explores the physics and chemistry of Nature with minimal reliance on mathematics Examines the structure and dynamics of the Universe and our Home Planet Earth Provides a detailed analysis of how humans as members of the Ecosphere have influenced and are continuing to influence the dynamics of events on the paludarium called Earth Presents underlying science of current political issues that shape the future of humankind Emphasizes how words and phrases and framing of issues can influence the opinions of members of society Makes extensive use of metaphors and everyday experiences to illustrate principles in science and social interactions Physics, Chemistry & Biology Solved Papers (2023-24 NEET/AIPMT) YCT Expert Team, 2023 24 NEET AIPMT Physics Chemistry Biology Solved Papers *NEET Exam Chemistry Question Bank* R P Meena, NEET Exam Chemistry Question Bank Topic wise and NCERT Based MCQs NEET Chemistry is considered to be the most scoring subject As it is one of the easiest scoring subjects it is often ignored and undermined a subject But if you want to get an edge over others here is a tip master NEET Chemistry concepts Chemistry demands the attention of students in understanding reactions and concrete basic understanding but once done it only gets easier from there NEET is amongst one the most prestigious medical entrance exams in India With just a few months left for the examination it becomes quite challenging for students to cover all the concepts included in the NEET syllabus thoroughly

However a proper study plan designed as per the latest examination pattern and the syllabus can help students to prepare all the important concepts in shorter time duration Given below are few useful tips that can assist the students in tackling multiple choice questions in NEET exam accurately In most of the multiple choice questions the options are designed in a very tricky and confusing manner In most of the cases all the given options seem to be correct in some aspect Therefore the students are advised to read the entire question very carefully Try to accumulate all the information provided in the question effectively because in some of the cases you can easily evaluate the Answers from the question itself If you are muddled by the given options then give each option a true and false test Instead of getting confused consider all the possibilities and neglect the incorrect options Hence in this way the most appropriate answer could be easily spotted Topic wise MCQs for NEET Chemistry Environmental Chemistry Organic Chemistry P Block Elements States of Matter Redox Reaction Chemical Equilibrium Thermodynamics Periodic Classification of Elements s Block Elements Chemical Bonding Molecular Structure Structure of Atom Some Basic Concepts of Chemistry Hydrocarbons Some Basic Concepts of Chemistry Solid State Structure of Atom Solutions Classification of Elements and Periodicity in Properties Electrochemistry Chemical Bonding and Molecular Structure Chemical Kinetics States of Matter Gases and Liquids Surface Chemistry Thermodynamics General Principles and Processes of Isolation of Elements Equilibrium Redox Reactions d and f Block Elements Hydrogen Coordination Compounds Haloalkanes and Haloarenes Some p Block Elements Alcohols Phenols and Ethers Organic Chemistry Some Basic Principles and Techniques Aldehyde Ketones and Carboxylic Acids Hydrocarbons Organic Compounds Containing Nitrogen Environmental Chemistry Biomolecules Polymers and Chemistry in Everyday Life Use a step wise approach to solve conceptual and complex questions Several times Matching type Questions are asked where the students are required to find the mismatched or the correctly matched option Some of the questions asked in the NEET exam are entirely memory based therefore the students are advised to memorize the common names scientific names concepts and important definitions Around 40% of the questions asked in the NEET exam are application based Therefore students need to focus more on the concepts along with its applications in order to score well in the examination Chemistry topics broadly can be grouped into Organic Inorganic and Physical chemistry Organic Chemistry is a vast section and holds the maximum weightage Its concepts and principles are important One needs to be conversant with properties of groups conversions recognizing organic compounds etc Interlinking between different chapters from organic chemistry can be of great help The students must primarily focus on reading NCERT textbooks Several times the questions asked in NEET exam are taken directly from the NCERT textbooks Initially avoid answering those questions for which you are not confident because your wrong answer may reduce your final score In order to utilize your time appropriately divide the three hours of examination time as per your comfort among Physics Chemistry and Biology Initially focus on attempting all easy questions and later on pick the difficult ones By this way your confidence will be elevated and you will also get more time to answer hard questions Practice previous

years question papers mock tests and sample papers to get an idea on how to answer MCQ questions efficiently Preparing at an early stage is what an MCQ exam requires Avoid guesswork for negative marking questions as they might lower your final score These tips can be very helpful for students to answer difficult and brain teaser questions Prior preparations and practice are mandatory aspects of any examination Hence to crack highly competitive examination like NEET it is mandatory for students to prepare well and acquire the skills to tackle multiple choice questions effectively Rather than just following mere guesswork the aspirants can focus on the tips discussed to tackle Multiple Choice Questions in NEET in the right manner

A Working Method Approach for Introductory Physical Chemistry Calculations Brian Murphy, Clair Murphy, Brian J. Hathaway, 1997 *A Working Method Approach for Introductory Physical Chemistry Calculations* is a concise inexpensive introduction to first year chemistry that is aimed at students who are weak in chemistry or have no chemistry on entry to university Such students usually find physical chemistry the most difficult part of the chemistry course and within this section numerical problem solving is an additional difficulty The text should also be invaluable to first year intending chemists This text provides an introduction to physical chemistry and the gas laws followed by chapters on thermodynamics chemical equilibrium electrochemistry and chemical kinetics Each section involves a brief introduction followed by a representative examination question which is broken down into a proposed working method Both short multiple choice questions and related full examination type questions are included This book will prove invaluable to students who need encouragement in a logical approach to problem solving in physical chemistry teaching them to think for themselves when faced with a problem

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principles The content is meticulously curated from various reputable chemistry textbooks and aligned with the syllabi of major competitive examinations Each question is designed not only to test knowledge but also to foster critical thinking and problem solving skills The questions and answers included in this book cover a wide array of topics from the mole concept and stoichiometry to chemical equilibrium and thermodynamics Each section is structured to build upon previous knowledge ensuring a progressive learning experience In addition to the theoretical aspects the book emphasizes practical applications helping students understand how chemistry relates to real world scenarios Whether you are a high school student preparing for entrance exams a college student revising fundamental concepts or a lifelong learner with an interest in chemistry this book serves as a valuable guide The aim is to facilitate a deeper understanding of chemistry empowering you to tackle challenges confidently and excel in your examinations As you navigate through the questions and answers I encourage you to engage actively with the material Consider each problem attempt to solve it independently and reflect on the explanations provided This interactive approach will enhance retention and understanding making your study sessions more productive I hope this book inspires a passion for chemistry and equips you with the tools needed to achieve your academic goals Remember the journey of learning is just as important as the destination and every question answered brings you one step closer to mastering this fascinating subject

MCAT General Chemistry Review The Princeton Review, 2015-03-17
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