

Elementary Chemicals I: Elemental Ions

Order is arbitrary; table is complete and stable before

Element Name	Element Symbol	Ion Name	Ion Formula
1. sodium	Na	sodium ion	Na^+
2. bromine	Br	bromide	Br^-
3. magnesium	Mg	magnesium ion	Mg^{2+}
4. chlorine	Cl	chloride	Cl^-
5. oxygen	O	oxide	O^{2-}
6. boron	B	boron ion	B^{3+}
7. lithium	Li	lithium ion	Li^+
8. neon	Ne	no ion	none
9. phosphorus	P	phosphide	P^{3-}
10. aluminum	Al	aluminum ion	Al^{3+}
11. calcium	Ca	calcium ion	Ca^{2+}
12. iodine	I	iodide	I^-
13. nitrogen	N	nitride	N^{3-}
14. cesium	Cs	cesium ion	Cs^+
15. sulfur	S	sulfide	S^{2-}
16. fluorine	F	fluoride	F^-
17. potassium	K	potassium ion	K^+
18. beryllium	Be	beryllium ion	Be^{2+}
19. hydrogen	H	hydrogen ion	H^+
20. helium	He	no ion	none

Monatomic Ions Worksheet 1 Answer Key

M Lipman

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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