



Density Calculations Worksheet I

$$\text{density} = \frac{\text{mass}}{\text{volume}}$$

UNITS OF DENSITY
solids (g/cm³) liquids (g/mL)

1. Find the unknown quantity:

a) $d = 3 \text{ g/mL}$
 $V = 100 \text{ mL}$
 $M = ?$
 $m = d \times v$
 $= (3 \text{ g/mL})(100 \text{ mL})$
 $= 3 \times 10^2 \text{ g}$

b) $d = ?$
 $V = 950 \text{ mL}$
 $M = 95 \text{ g}$
 $d = \frac{m}{V}$
 $= \frac{95 \text{ g}}{950 \text{ mL}} = 0.10 \text{ g/mL}$

c) $d = 0.5 \text{ g/cm}^3$
 $V = ?$
 $M = 20 \text{ g}$
 $V = \frac{m}{d}$
 $= \frac{20 \text{ g}}{0.5 \text{ g/cm}^3} = 4 \times 10^1 \text{ cm}^3$

2. Find the unknown quantity (CONVERT FIRST to g or mL)

a) $d = 24 \text{ g/mL}$
 $V = 1.2 \text{ L} = \underline{1200} \text{ mL}$
 $M = ?$
 $m = d \times v$
 $= (24 \text{ g/mL})(1200 \text{ mL})$
 $= 2.9 \times 10^4 \text{ g}$

b) $d = ?$
 $V = 100 \text{ mL}$
 $M = 1.5 \text{ kg} = \underline{1500} \text{ g}$
 $d = \frac{m}{V}$
 $= \frac{1500 \text{ g}}{100 \text{ mL}} = 15.0 \text{ g/mL}$

c) $d = ?$
 $V = 0.52 \text{ L} = \underline{520} \text{ mL}$
 $M = 500 \text{ mg} = \underline{0.500} \text{ g}$
 $d = \frac{m}{V}$
 $= \frac{0.500 \text{ g}}{520 \text{ mL}} = 9.62 \times 10^{-4} \text{ g/mL}$

WORD PROBLEMS

1. A block of aluminum occupies a volume of 15.0 mL and weighs 40.5 g. What is its density?

$V = 15.0 \text{ mL}$
 $m = 40.5 \text{ g}$
 $d = ?$
 $d = \frac{m}{V}$
 $= \frac{40.5 \text{ g}}{15.0 \text{ mL}}$

$d = 2.70 \text{ g/mL}$

2. Mercury metal is poured into a graduated cylinder that holds exactly 22.5 mL. The mercury used to fill the cylinder weighs 306.0 g. From this information, calculate the density of mercury.

$V = 22.5 \text{ mL}$
 $m = 306.0 \text{ g}$
 $d = ?$

$d = \frac{m}{V} = \frac{306.0 \text{ g}}{22.5 \text{ mL}} = 13.6 \text{ g/mL}$

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Table of Contents M Elkhatib Calculations Worksheet Answers

1. Understanding the eBook M Elkhatib Calculations Worksheet Answers
 - The Rise of Digital Reading M Elkhatib Calculations Worksheet Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying M Elkhatib Calculations Worksheet Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an M Elkhatib Calculations Worksheet Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from M Elkhatib Calculations Worksheet Answers
 - Personalized Recommendations
 - M Elkhatib Calculations Worksheet Answers User Reviews and Ratings
 - M Elkhatib Calculations Worksheet Answers and Bestseller Lists

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 - M Elkhatib Calculations Worksheet Answers eBook Subscription Services
 - M Elkhatib Calculations Worksheet Answers Budget-Friendly Options
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 - ePub, PDF, MOBI, and More
 - M Elkhatib Calculations Worksheet Answers Compatibility with Devices
 - M Elkhatib Calculations Worksheet Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of M Elkhatib Calculations Worksheet Answers
 - Highlighting and Note-Taking M Elkhatib Calculations Worksheet Answers
 - Interactive Elements M Elkhatib Calculations Worksheet Answers
8. Staying Engaged with M Elkhatib Calculations Worksheet Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers M Elkhatib Calculations Worksheet Answers
9. Balancing eBooks and Physical Books M Elkhatib Calculations Worksheet Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection M Elkhatib Calculations Worksheet Answers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine M Elkhatib Calculations Worksheet Answers
 - Setting Reading Goals M Elkhatib Calculations Worksheet Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of M Elkhatib Calculations Worksheet Answers
 - Fact-Checking eBook Content of M Elkhatib Calculations Worksheet Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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