

ಘಟಕ 1 : ಸಿನಿಮಾ / ಮನೋರಂಜನೆ

1. ಉತ್ತರ ಕುಮಾರ (ಕಾವ್ಯ)

- ಕುಮಾರವ್ಯಾಸ

ಕವಿ ಪರಿಚಯ :

ಕನ್ನಡದ ಅತ್ಯುನ್ನತ ಕವಿಗಳಲ್ಲಿ ಕುಮಾರವ್ಯಾಸ ಒಬ್ಬರಾಗಿದ್ದಾರೆ. ಕುಮಾರವ್ಯಾಸನ ಮೂಲ ಹೆಸರು ಗದುಗಿನ ನಾರಾಯಣಪ್ಪ. ಈತನ ಕಾಲ ಕ್ರಿ.ಶ. 1400 ಈತನ ಊರು ಕೋಳಿವಾಡ. ಸಂಪ್ರದಾಯದ ಕವಿಯಾದ ನಾರಣಪ್ಪ ಗದುಗಿನ ವೀರನಾರಾಯಣನ ರಾಮ ಭಕ್ತನಾಗಿದ್ದನು.

ಕುಮಾರವ್ಯಾಸನನ್ನು ರೂಪಕ ಸಾಮ್ರಾಜ್ಯ ಚಕ್ರವರ್ತಿ ಎಂದು ಕರೆಯಲಾಗಿದೆ. ಕುಮಾರವ್ಯಾಸನು ಕರ್ಣಾಟ ಭಾರತ ಕಥಾಮಂಜರಿ ಕೃತಿಯನ್ನು ರಚಿಸಿದ್ದಾನೆ. ಕರ್ಣಾಟ ಭಾರತ ಕಥಾಮಂಜರಿಗೆ ಕುಮಾರವ್ಯಾಸ ಭಾರತ, ಗದುಗಿನ ಭಾರತ ಎಂಬ ಹೆಸರುಗಳೂ ಇವೆ.

ಆಶಯ :

ಮಹಾಭಾರತದ ಕತೆಯಲ್ಲಿ ಬರುವ ವಿರಾಟರಾಯನ ಮಗನಾದ ಉತ್ತರಕುಮಾರ ಹುಟ್ಟಿನಿಂದಲೂ ಕ್ಷತ್ರಿಯನಾಗಿದ್ದೂ ಆತ ಅಂಜುಬುರುಕನಾಗಿದ್ದಾನೆ. ಮಹಿಳೆಯರ/ ಹೆಂಗಳೆಯರ ಮುಂದೆ ಪೌರುಷವನ್ನು ತೋರಿಸಿಕೊಳ್ಳುತ್ತ ಪುರುಷರೊಂದಿಗೆ ಹೋರಾಟ ನಡೆಸಲು ಹೆದರಿಕೊಳ್ಳುವ ಸ್ವಭಾವದವನಾಗಿದ್ದಾನೆ. ಉತ್ತರಕುಮಾರ ಹಾಸ್ಯರಸ ಪ್ರತಿಪಾದನೆಗಾಗಿಯೇ ಕುಮಾರವ್ಯಾಸ-ಉತ್ತರಕುಮಾರನ ಪಾತ್ರವನ್ನು ಸೃಷ್ಟಿಸಿದ್ದಾನೆ.

ಸಾರಾಂಶ :

- 1) ಕೀಳು ಜನಮೇಜಯ ಸುಯೋಧನ
ನಾಳು ಮುಕ್ತಿತ್ತು ಪುರುಗಳನು ಮೇ
ಲಾಳು ಕವಿದುರು ಭೀಷ್ಮ ಕರ್ಣ ದ್ರೋಣ ಮೊದಲಾಗಿ
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 $\text{NaCl} + \text{F}_2 \rightarrow \text{NaF} + \text{Cl}_2$
 $2 \text{H}_2 + \text{O}_2 \rightarrow 2 \text{H}_2\text{O}$
 $\text{Pb} + \text{HCl} \rightarrow \text{PbCl}_2 + \text{H}_2$
 $\text{Al} + \text{Br}_3 \rightarrow \text{AlBr}_3$
 $3 \text{K}_2\text{SO}_4 \rightarrow 6 \text{KBr} + \text{Al}_2\text{SO}_4 + 3 \text{CH}_4 + 2 \text{O}_2$
 $\text{CO}_2 + 2 \text{H}_2\text{O} \rightarrow \text{C}_3\text{H}_8 + 5 \text{O}_2$
 $3 \text{CO}_2 + 4 \text{H}_2\text{O} \rightarrow \text{C}_8\text{H}_{18} + 25 \text{O}_2$
 $\text{Al}_6\text{CO}_2 + 18 \text{H}_2\text{O}$

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 butane C_4H_{10} reacts with diatomic oxygen gas to yield gaseous carbon dioxide and water vapor

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 $\text{Na}_3\text{PO}_4 + 3\text{KOH} \rightarrow 3\text{NaOH} + \text{K}_3\text{PO}_4 + 1\text{MgF}_2 + 1\text{Li}_2\text{CO}_3 + 1\text{MgCO}_3 + 1\text{LiF} + 4\text{H}_2\text{O} + 2\text{P}_2\text{O}_5 + 3\text{RbNO}_3 + 1\text{BeF}_2 + 1\text{BeNO}_3 + 2\text{RbF}$

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 $4\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$
 $5\text{Pb} + 2\text{HCl} \rightarrow 2\text{H}_2 + \text{PbCl}_2$
 $6\text{AlBr}_3 + 3\text{K}_2\text{SO}_4 \rightarrow 4\text{KBr} + \text{Al}_2\text{SO}_4 + 3\text{H}_4$
 $2\text{CH}_4 + \text{O}_2 \rightarrow 2\text{H}_2\text{O} + 8\text{H}_2\text{O} + 1\text{C}_3\text{H}_8 + 5\text{O}_2 + 3\text{CO}_2 + 4\text{H}_2$

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 $1\text{O}_2 + 3\text{HNO}_3 \rightarrow 3\text{H}_2\text{O} + 2\text{O}_2 + 3\text{H}_2\text{O} + 2\text{H}_2\text{O} + 4\text{RbH} + 2\text{ORbOH} + 5\text{Cl}_2 + 2\text{KBr} + \text{KCl} + \text{Br}_2 + 6\text{N}_2 + 2\text{H}_2 + 3\text{NH}_3 + 7\text{N}_2\text{F}_3 + \text{Ca} + \text{Nd} + \text{CaF}_2 + 8\text{CH}_4 + 2\text{O}_2 + \text{CO}_2 + 9\text{C}_2\text{H}_5\text{OH}$

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