

Question Number	Acceptable Answers	Mark
24(b)(i)	<p>EITHER</p> $\begin{array}{rcl} \text{C} & : & \text{H} & : & \text{O} \\ \text{mol } 64.9 & : & 13.5 & : & 21.6 \\ & 12 & 1 & 16 \\ = & 5.408 & : & 13.5 & : & 1.35 \\ = & 4.006 & : & 10 & : & 1 \\ = & 4 & : & 10 & : & 1 \end{array}$ <p>use of 74 to show molecular formula is $\text{C}_4\text{H}_{10}\text{O}$ eg M_r is $(4 \times 12) + (10 \times 1) + 16 = 74$</p> <p>OR</p> <p>C atoms = $\frac{64.9 \times 74}{100 \times 12} = 4$</p> <p>H atoms = $\frac{13.5 \times 74}{100 \times 1} = 10$</p> <p>O atoms = $\frac{21.6 \times 74}{100 \times 16} = 1$</p> <p>This may be done in 2 steps eg C $\frac{64.9 \times 74}{100} = 48$ $\frac{48}{12} = 4$</p> <p>All 3 correct scores 2 Any 2 correct scores 1</p> <p>OR</p> <p>% C = $\frac{48 \times 100}{74} = 64.9$</p> <p>% H = $\frac{10 \times 100}{74} = 13.5$</p> <p>% O = $\frac{16 \times 100}{74} = 21.6$</p> <p>All 3 correct scores (2) Any 2 correct scores (1)</p>	(2)

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Sabine Zange



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