



# Hp Laserjet 2200dn Service Manual

**Hewlett-Packard Company**



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web solution  $v_1 = 250 \text{ l}$   $v_2 = 273 \text{ 30 303 k}$   $t_2 = 273 \text{ 70 343 k}$  we will be applying charles law as pressure is constant here  $v_1 \cdot t_1 = v_2 \cdot t_2$  or  $v_2 = v_1 \cdot t_1 \cdot x \cdot t_2 = 250 \text{ l} \cdot 303 \text{ k} \cdot x \cdot 343 \text{ k} = 283 \text{ l}$  at 90 c a helium sample has a volume of 500 ml determine the temperature at which the volume of the liquid will become 240 ml

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