

Study Guide for Chemistry EOC Assessment

Your test will consist of multiple-choice questions over the following topics.

1. Metric units for mass, length, volume
2. Metric to metric conversions (K H Da m/l/g D C M)
3. Know how to put numbers in scientific notation and take them out.
4. Significant Digit rules
5. Difference between chemical and physical properties and changes. There will be multiple questions on this.
6. Ways to tell if a chemical change (chemical reaction) is taking place.
7. Pure substances and mixtures
 - a. heterogeneous mixture
 - b. homogeneous mixture
8. Density & how it relates to floating and sinking
9. Use the equation for density
$$\text{Density} = \frac{\text{Mass}}{\text{volume}}$$

(g/ml) (g) (ml)
10. Know the phases of matter and the characteristics of each phase.
11. Law of Conservation of Matter
12. What is a mole in chemistry?
13. Convert grams to moles, and moles to grams
14. Define an atom
15. Structure of an atom and where each subatomic particle is located.
 - a. Nucleus (contains protons and neutrons)
 - b. Protons (positive)
 - c. Neutrons (neutral)
 - d. Electrons (negative)
16. Isotopes
17. Elements
18. If given atomic # and mass, find the number of protons and neutrons in an atom.
19. Find the number of valence electrons, and tell what it says about reactivity.
20. Periodic Table
 - a. Know about the families
 - b. Group vs. period
21. Know about the properties of metals and nonmetals.
 - a. Ductile, malleable, conductors
22. Ionic & Covalent Compounds

How do they form? (Transfer or gain electrons)

Characteristics of Each
23. Be able to name ionic compounds
24. Predict oxidation numbers
25. Chemical Equations:

Know the parts of an equation

5 main types of reactions-know about them and be able to identify them (Synthesis, Decomposition, Single Displacement, Double Displacement, Combustion)
26. Acids & Bases-

Macro and Microscopic properties of each

What ions do they form, how do they affect litmus, what is their pH range?

Identify examples of each- Mono, di, and polyprotic acids

Tell if they are strong or weak

Acid-base indicator

How To Study For The Tn Chemistry Eoc Exam

SJ Ball



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Table of Contents How To Study For The Tn Chemistry Eoc Exam

1. Understanding the eBook How To Study For The Tn Chemistry Eoc Exam
 - The Rise of Digital Reading How To Study For The Tn Chemistry Eoc Exam
 - Advantages of eBooks Over Traditional Books
2. Identifying How To Study For The Tn Chemistry Eoc Exam
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an How To Study For The Tn Chemistry Eoc Exam
 - User-Friendly Interface
4. Exploring eBook Recommendations from How To Study For The Tn Chemistry Eoc Exam
 - Personalized Recommendations
 - How To Study For The Tn Chemistry Eoc Exam User Reviews and Ratings
 - How To Study For The Tn Chemistry Eoc Exam and Bestseller Lists
5. Accessing How To Study For The Tn Chemistry Eoc Exam Free and Paid eBooks
 - How To Study For The Tn Chemistry Eoc Exam Public Domain eBooks
 - How To Study For The Tn Chemistry Eoc Exam eBook Subscription Services
 - How To Study For The Tn Chemistry Eoc Exam Budget-Friendly Options
6. Navigating How To Study For The Tn Chemistry Eoc Exam eBook Formats

- ePub, PDF, MOBI, and More
- How To Study For The Tn Chemistry Eoc Exam Compatibility with Devices
- How To Study For The Tn Chemistry Eoc Exam Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of How To Study For The Tn Chemistry Eoc Exam
 - Highlighting and Note-Taking How To Study For The Tn Chemistry Eoc Exam
 - Interactive Elements How To Study For The Tn Chemistry Eoc Exam
- 8. Staying Engaged with How To Study For The Tn Chemistry Eoc Exam
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers How To Study For The Tn Chemistry Eoc Exam
- 9. Balancing eBooks and Physical Books How To Study For The Tn Chemistry Eoc Exam
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection How To Study For The Tn Chemistry Eoc Exam
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine How To Study For The Tn Chemistry Eoc Exam
 - Setting Reading Goals How To Study For The Tn Chemistry Eoc Exam
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of How To Study For The Tn Chemistry Eoc Exam
 - Fact-Checking eBook Content of How To Study For The Tn Chemistry Eoc Exam
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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