

## Chemistry – Mock Test Paper

[Time:  $1\frac{1}{2}$  hrs]

[M. Marks : 80]

Answers to this paper must be written on the paper provided separately.

You will not be allowed to write during the first 15 minutes.

This time is to be spent in reading the Question Paper.

The time given at the head of this paper is the time allowed for writing the answers.

Section I is compulsory. Attempt any four questions from Section II.

### Section I (40 Marks)

Attempt all questions from this section

#### Question 1.

(a) Choose the most appropriate answer.

- (i) The element with largest size in second period is :  
 A. Lithium                      B. Oxygen                      C. Fluorine                      D. Carbon
- (ii) Which is the element which belongs to group 13 from :  
 A. Beryllium                      B. Boron                      C. Aluminium                      D. Carbon
- (iii) The most strong alkali is  
 A. Lithium hydroxide                      B. Potassium hydroxide  
 C. Sodium hydroxide                      D. Rubidium hydroxide
- (iv) Lanthanides is a group of elements outside the Periodic Table having atomic number from  
 A. 57 to 71                      B. 89 to 103                      C. 72 to 86                      D. 104 to 118
- (v) The elements with lowest first ionisation energy is  
 A. Sodium                      B. Caesium                      C. Barium                      D. Magnesium
- (vi) When equal volumes of hydrogen gas and chlorine gas are exposed to diffused sunlight the reaction :  
 A. does not take place                      B. takes place at moderate speed  
 C. is explosive in nature                      D. none of the above.
- (vii) In laboratory ammonia gas prepared by heating a mixture of :  
 A.  $\text{NH}_4\text{Cl}$  and  $\text{Ca}(\text{OH})_2$                       B.  $(\text{NH}_4)_2\text{SO}_4$  and  $\text{Ca}(\text{OH})_2$   
 C.  $\text{NH}_4\text{NO}_3$  and  $\text{KOH}$                       D.  $\text{NH}_4\text{NO}_3$  and  $\text{Ca}(\text{OH})_2$
- (viii) The compound which will liberate carbon dioxide on treating with dilute sulphuric acid is :  
 A.  $\text{MgCl}_2$                       B.  $\text{Na}_2\text{SO}_4$                       C.  $\text{Na}_2\text{CO}_3$                       D.  $\text{Na}_2\text{SO}_4$
- (ix) Which is not a member of alkanolic series?  
 A.  $\text{HCOOH}$                       B.  $\text{C}_2\text{H}_5\text{COOH}$                       C.  $\text{HCHO}$                       D.  $\text{CH}_3\text{COOH}$
- (x) Ethyne can be converted into ethyne ozonide by treating with :  
 A. Hydrogen peroxide                      B. Ozone  
 C. Oxygen                      D. Conc.  $\text{HNO}_3$

[10]

# Grade Chemistry Exam Papers

**Mark Steyvers**



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### **Table of Contents Grade Chemistry Exam Papers**

1. Understanding the eBook Grade Chemistry Exam Papers
  - The Rise of Digital Reading Grade Chemistry Exam Papers
  - Advantages of eBooks Over Traditional Books
2. Identifying Grade Chemistry Exam Papers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Grade Chemistry Exam Papers
  - User-Friendly Interface
4. Exploring eBook Recommendations from Grade Chemistry Exam Papers
  - Personalized Recommendations
  - Grade Chemistry Exam Papers User Reviews and Ratings
  - Grade Chemistry Exam Papers and Bestseller Lists

- 
5. Accessing Grade Chemistry Exam Papers Free and Paid eBooks
    - Grade Chemistry Exam Papers Public Domain eBooks
    - Grade Chemistry Exam Papers eBook Subscription Services
    - Grade Chemistry Exam Papers Budget-Friendly Options
  6. Navigating Grade Chemistry Exam Papers eBook Formats
    - ePub, PDF, MOBI, and More
    - Grade Chemistry Exam Papers Compatibility with Devices
    - Grade Chemistry Exam Papers Enhanced eBook Features
  7. Enhancing Your Reading Experience
    - Adjustable Fonts and Text Sizes of Grade Chemistry Exam Papers
    - Highlighting and Note-Taking Grade Chemistry Exam Papers
    - Interactive Elements Grade Chemistry Exam Papers
  8. Staying Engaged with Grade Chemistry Exam Papers
    - Joining Online Reading Communities
    - Participating in Virtual Book Clubs
    - Following Authors and Publishers Grade Chemistry Exam Papers
  9. Balancing eBooks and Physical Books Grade Chemistry Exam Papers
    - Benefits of a Digital Library
    - Creating a Diverse Reading Collection Grade Chemistry Exam Papers
  10. Overcoming Reading Challenges
    - Dealing with Digital Eye Strain
    - Minimizing Distractions
    - Managing Screen Time
  11. Cultivating a Reading Routine Grade Chemistry Exam Papers
    - Setting Reading Goals Grade Chemistry Exam Papers
    - Carving Out Dedicated Reading Time
  12. Sourcing Reliable Information of Grade Chemistry Exam Papers
    - Fact-Checking eBook Content of Grade Chemistry Exam Papers
    - Distinguishing Credible Sources
  13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

#### 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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