

A crystalline solid is formed when 2 or more atoms exchange valence electrons to increase stability.

Ionic bond is formed when valence electrons are transferred from one atom to another.

cations = atoms lose electrons to become [+] charged ions.

atoms gain electrons to become [-] charged.

In ionic compounds that ions are arranged in a crystal lattice. Strong forces hold the ions together.

Properties of ionic compounds:

- = high melting and boiling points
- = hard & not easily crushed
- = conduct electricity when melted or dissolved because the ions are free to move

covalent bond = electrons are shared. Forming molecules

Covalent compounds have weak forces holding the molecules together.

Properties of covalent compounds:

- low melting and boiling points
- Many covalent compounds are volatile liquids or gases.
- soft, easier to crush
- are not conductors of electricity
- electronegativity = property that tells how strong an

Gbp Episode 6 answers

S Ben Porath



Gbp Episode 6 answers:

Unveiling the Energy of Verbal Artistry: An Mental Sojourn through **Gbp Episode 6answers**

In a world inundated with monitors and the cacophony of quick interaction, the profound power and mental resonance of verbal beauty usually disappear into obscurity, eclipsed by the continuous onslaught of noise and distractions. However, nestled within the musical pages of **Gbp Episode 6answers**, a captivating function of fictional brilliance that impulses with fresh thoughts, lies an unique trip waiting to be embarked upon. Published by a virtuoso wordsmith, this interesting opus courses viewers on a psychological odyssey, gently revealing the latent possible and profound affect embedded within the elaborate web of language. Within the heart-wrenching expanse with this evocative examination, we shall embark upon an introspective exploration of the book is main styles, dissect their fascinating publishing fashion, and immerse ourselves in the indelible impact it leaves upon the depths of readers souls.

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