

Mole and Stoichiometry Mini-Lab

7. Write and balance the equation for the reaction between aluminum and water to produce aluminum oxide and hydrogen gas.



8. Using the moles of the aluminum can in #1 as the GIVEN, how many moles of hydrogen gas can be produced? Assume water is available in excess. UNKNOWN: mol H_2 GIVEN: mol Al (answer to #1)

9. Using the moles of water from #4 as the GIVEN with aluminum in excess, calculate the number of moles of aluminum oxide that can be produced.



Mini Lab 1 apply Stoichiometry

Yijin Wang



Mini Lab 1 apply Stoichiometry:

Reviewing **Mini Lab 1apply Stoichiometry**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is actually astonishing. Within the pages of "**Mini Lab 1apply Stoichiometry**," an enthralling opus penned by a highly acclaimed wordsmith, readers embark on an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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