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they are the same as basic solutions

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thus ruling out that S^{2-} is the weakest base

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web solutions manual acids i and bases section 18 1 introduction to acids and bases pages 634 643 practice problems pages 635 640 write balanced equations for reactions between the following aluminum and sulfuric acid $2\text{Al} + 3\text{H}_2\text{SO}_4 \rightarrow \text{Al}_2(\text{SO}_4)_3 + 3\text{H}_2$ calcium carbonate and hydrobromic acid