

TEXT STRUCTURE

Cause & Effect

NAME: _____ DATE: _____

CAUSE & EFFECT: Passage

Directions: Read the passage below about motion sickness. Underline each cause that you find. Circle each effect that you find.

The Tummy Twist

Have you ever been on a boat, a car, or an amusement park ride and started to feel kind of funny in your stomach? This is called motion sickness. It's like a little puzzle that happens in our bodies.

Imagine you're sitting in a car or on a boat. Your eyes look at things around you, like the road or the water. At the same time, your inner ears, which help you keep your balance, feel the movement. Sometimes, these two things don't match. This can happen when you read a book while riding in a car. Your eyes see the book staying still, but your ears feel the car moving. This mix-up confuses your brain. That's when motion sickness can sneak in.

When your brain gets confused, it might send signals to your stomach. This can make you feel sick. You might notice more spit in your mouth, your stomach might feel upset, and you might even get sweaty. It's not a great feeling, but it's your body's way of trying to figure things out.

But don't worry, there are things you can do to help prevent or ease motion sickness. One trick is to look outside, straight ahead. This helps your eyes and inner ears match up the movements better. You can also try to keep still and not read or play on screens while you're moving. Keeping your mind busy with fun things, like talking to your friends or singing songs, can help distract you from that funny feeling.

If you're going on a car ride, it's a good idea to have some fresh air by opening a window. Chewing gum or snacking on crackers might help too. And if you're at an amusement park and want to go on those exciting rides, try starting with the slower ones. Then, move to the faster ones. This gives your body a chance to get used to the motion.

NAME: _____ DATE: _____

CAUSE & EFFECT: Passage

Directions: Complete the graphic organizer below using the information you read in *The Tummy Twist*. Read passage. Then, answer the questions in complete sentences.

Cause	Cause
↓	↓
Effect	Effect

1. Motion sickness?

2. Motion sickness happens?



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for the nitration 4 01 ml of nitric acid 4 00 ml of sulfuric acid and 4 56 ml bromobenzene were used to complete the reaction with these measurements bromobenzene was found to be the limiting reagent since only 4 56 ml of bromobenzene was used the amount of product predicted to be recovered was 8 7533 g

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the aim of the experiment is to determine and study electrophilic aromatic substitution reaction eas particularly nitration of methyl benzoate go give methyl m nitrobenzoate thus determine the weight melting point and percentage yield of the pure methyl m nitrobenzoate abstract

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in this experiment we will carry out the nitration of bromobenzene bromine substituted on an aromatic ring is an ortho para directing group it will direct an incoming electrophile to the ortho and para positions it is also deactivating meaning that its reactions will be slower than the identical reactions of benzene

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section 20711 lab experiment nitration of bromobenzene introduction the chemical reaction of nitration consists of a nitro group being added to or substituted in a molecule nitration can basically be carried out by a mixture of concentrated nitric acid and sulfuric acid this mixture is useful to obtain the active nitronium ion

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1 go to the restroom and wash the affected area with lots of soap and water the soap in the lab is not suitable for this purpose 2 return to the lab and apply mineral oil to the affected area 3 the summary to this warning is not to touch any part of your body while performing this experiment if you must leave the lab for any reason 1

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in the nitration of bromobenzene a mixture of sulfuric acid and nitric acid react with one another nitric acid the weaker of the two acids will serve as base to form nitronium ion bromobenzene will attack the nitronium ion by nucleophilic attack

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then actual is your actual findings yield actual theoretical $\times 100$ what are the possible side reactions of nitration of bromobenzene dinitration can occur at ortho para or at both ortho positions occurs in overheating meta product happens very rarely because bromine is an ortho para director

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the soap in the lab is not suitable for this purpose 2 return to the lab and apply mineral oil to the affected area theory and explanation of the nitration of bromobenzene if the temperature of the reaction mixture exceeds 60 o

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the bromine substituent in bromobenzene is an ortho para director the para position is favored due to steric effects in the reaction that takes place in this lab a nitro group is substituted onto bromobenzene at the para position to form 1 bromo 4 nitrobenzene

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study with quizlet and memorize flashcards containing terms like reacting sulfuric and nitric acid together produces the nitronium ion alone nitric acid is not a strong enough electrophile to react with bromobenzene sulfuric acid is added to the

nitric acid the reverse rxn can overheat and boil easily includes a claisen adaptor and

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over 5 minutes slowly pipet this bromobenzene into the icy cold nitration mixture hno 3 h 2 so 4 in the 125 ml flask and stir continue to stir the reaction mixture in the ice bath for 30 minutes do not heat in a new 250 ml erlenmeyer

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on the right the arenium cation has a nitro set with incredibly small 0 6 angstrom bond distances this will result in an optimization failure as the atoms are even close for the programme to accomodate their electrons demonstration of creating the p arenium cation by to nitration of bromobenzene

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the purpose of this lab was to observe where substituents are added on a benzene ring by way of electrophilic aromatic substitution the electrophilic source in the nitration of bromobenzene comes from a nitronium ion no2 which acts as the electrophilic species

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nitration of bromobenzene is an electrophilic aromatic substitution reaction the electrophile will be formed from 4 0ml of nitric acid and 4 0ml of sulfuric acid and will then react with 4 5ml of bromobenzene this will produce o bromonitrobenzene and p bromonitrobenzene

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