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Metro Pcs User Guide

Lingsheng Yao



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$$\text{H}_2(\text{g}) + \text{I}_2(\text{g}) \rightleftharpoons 2\text{HI}(\text{g})$$
at 425°C an equilibrium mixture has the following concentrations
 $[\text{H}_2] = 0.1 \times 10^{-2} \text{ mol l}^{-1}$, $[\text{I}_2] = 2.25 \times 10^{-3} \text{ mol l}^{-1}$, $[\text{HI}] = 2.49 \times 10^{-3} \text{ mol l}^{-1}$

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