

Marks: 225

Time: 3 hours

SECTION A

QUESTION 1

1.1.1. False ✓

1.1.2. False ✓

1.1.3. True ✓

1.1.4. False ✓

1.1.5. False ✓

1.1.6. True ✓

1.1.7. True ✓

1.1.8. False ✓

1.1.9. True ✓

1.1.10. True ✓

(10x1) (10)

1.2.

1.2.1. Coriolis ✓

1.2.2. Batholith ✓

1.2.3. Tropical easterlies ✓

1.2.4. Water table ✓

1.2.5. Sedimentary ✓

(5x1) (5)

1.3.

1.3.1. High pressure ✓

(1x1) (1)

1.3.2. Pressure is highest at the centre of the cell; ✓ ✓ pressure decreases outwards; ✓ ✓

1.3.3. the latitudinal position of the cell indicates it is high pressure. ✓ ✓ (any one)

(1x2) (2)

1.3.4. Cold front ✓

(1x1) (1)

1.3.5. Summer. ✓ High temperatures ✓ / position of the high pressure cells ✓ / no Kalahari anticyclone is present ✓ (any one correct reason)

(2x1) (2)

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JG Myers



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enthalpy change in kJ for this reaction assume the densities of the solutions are 1.00 g/mL and that

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web choose 1 answer the final temperature is closer to t_1 than to t_2 a the final temperature is closer to t_1 than to t_2 the final temperature is exactly halfway between t_1 and t_2 b the final temperature is exactly halfway between t_1 and t_2 the final temperature is closer to t_2 than to t_1 c

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web 6 a 218 g sample of steam at 121°C is cooled to ice at 14°C find the change in heat content of the system 7 in going from

ice at 34°C to steam at 138°C a sample of water absorbs $1.41 \times 10^5 \text{ J}$ find the mass of the sample answers $1.268 \times 10^4 \text{ g}$ $2.623 \times 10^4 \text{ g}$ $3.171 \times 10^4 \text{ g}$ $4.111 \times 10^5 \text{ g}$ $5.211 \times 10^5 \text{ g}$ 6

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